# The geology and geochemistry of the Strathy complex of north-east Sutherland, Scotland

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ABSTRACT. The Strathy complex is a fault-bounded block of the sub-Moine succession basement characterized by a marked aeromagnetic anomaly. XRF analyses of 94 rocks for 26 elements are used to illustrate the geochemistry of the complex. It comprises siliceous grey gneisses (50 analyses), subordinate amphibolite (13 analyses), hornblende gneiss (13 analyses), and ultramafites (1 analysis), with scapolite-diopside marble and calc-silicate rocks (3 analyses from Strathy, 14 analyses of other calc-silicates). Geochemically the Strathy gneisses are quite distinct from the adjacent Moine lithologies, being unusually low in K, Rb, La, and Ce, and high in SiO<sub>2</sub> and Na<sub>2</sub>O, whilst differences in immobile elements such as Ti and Zr preclude their derivation from a Moine source. The Strathy complex is unlike other high-grade gneiss terrains and apparently acquired its characteristic geochemical features by the removal of an anatectic melt under amphibolite-facies conditions leaving a refractory residue of quartz, plagioclase, Ca-poor amphibole, and garnet ± staurolite in siliceous gneisses. Subsequent influx of water, retrogression to biotite-bearing assemblages, production of sillimanite, and anatectic trondhjemitic melts took place during the maximum metamorphism of the surrounding Moine. The complex is considered to have originated as a dacitic supracrustal sequence with minor sediments and mafic-ultramafic intrusives. REE and Y abundances suggest a Proterozoic age, possibly as part of the Laxfordian cycle.

# Introduction and geological setting of the Strathy complex

The Strathy complex occurs as a fault-bounded block of sub-Moine basement (Harrison and Moorhouse, 1976) within the migmatitic gneisses and granites of the east Sutherland Moine. This eastern Moine is separated from the west Sutherland 'A' Mhoine-type' feldspathic psammites by a central belt (some 11 km wide) bounded by two complex tectonic discontinuities, the Naver and Meadie 'slide' zones (fig. 1). This belt comprises psammite and subordinate pelitic rocks and is dominated by six major tectonic inliers and numerous minor slices believed to be derived from the sub-Moine basement (Read, 1931; Winchester and Lambert, 1970; Moorhouse, 1976; Moorhouse and Harrison, 1976; Moorhouse and Moorhouse.

1977). As distinct from the Strathy complex, these tectonic inliers are folded sheets concordant with the regional foliation, and rocks adjacent to their boundaries commonly contain mineral fabrics characteristic of extremely high strain, taken to indicate their emplacement by a mechanism of low-angle ductile thrusting (sliding) quite different from the vertical shear-zones bounding the Strathy complex.

In the north, along the well-exposed coastal strip, the east Sutherland belt is divisible on lithological and structural grounds into four assemblages. (a) The Bettyhill assemblage of variably migmatized quartzitic and pelitic gneiss, amphibolite and several suites of concordant, and cross-cutting, often foliated granite, with thin, strongly reworked strips of the sub-Moine basement (fig. 1). (b) The Kirtomy assemblage of banded, migmatitic biotite gneisses with rarely foliated granites and subordinate amphibolite lies structurally above the Bettyhill assemblage and is separated from it by the Swordly slide which brings up slivers of sub-Moine basement. (c) The Strathy complex which apparently occurs faulted into the Kirtomy assemblage, the two being juxtaposed east of Ardmore Point (fig. 1), with the eastern boundary on the north coast obscured by the Strathy Bay Old Red Sandstone which is possibly infilling a fault valley. (d) The Portskerra assemblage of quartitic gneiss, abundant granite, and occasional amphibolites which is reminiscent of the Bettyhill assemblage.

## The Strathy complex

The Strathy complex (Harrison and Moorhouse, 1976) comprises mainly K-feldspar-free siliceous grey gneisses, subordinate hornblende gneiss, amphibolite, very rare ultramafites, and marble with calc-silicates, intimately associated with later strings, sheets, veins, and larger bodies of concordant and cross-cutting, largely unfoliated trondhjemite.

These rocks are conspicuously different, both lithologically and geochemically (see later), from

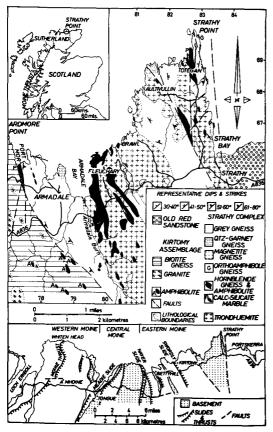


Fig. 1. Geological map of the Strathy complex and its tectonic setting within the north Sutherland Moine.

the undoubted Moine of north Sutherland. That this difference is of a fundamental nature is confirmed by a study of the aeromagnetic map of Great Britain (IGS, 1972) where the Strathy complex is clearly defined by a strong positive anomaly which closely follows the mapped boundaries of the complex. Furthermore the anomaly suggests an off-shore extension of the complex approximately equal in area to that mapped onshore. The magnetic anomaly is abruptly terminated some 9 km offshore, possibly by an ENE-trending fault. The close spacing of anomaly contours along the boundaries of the complex is taken as further confirmation of an origin as a fault-bounded block of lower crustal material.

## Structure of the complex

The Strathy complex is characterized by an apparent simplicity of structure, as distinct from the involved sequence of at least five phases of

folding recognized in the Bettyhill and Kirtomy assemblages (Moorhouse, 1979; Moorhouse and Moorhouse, 1979b, and in prep.) and minor folds and refolds are only sparingly developed in the poorly banded rocks within the complex. The regional foliation of the complex outlines large, rather open NNW-trending folds (fig. 1) believed to be equivalent to late folds elsewhere in north-east Sutherland.

The western boundary of the complex is well exposed on the coast at Port Mor (fig. 1), and a marked escarpment can be traced south-eastwards for 8 km before running into extensive peat bog. At Port Mor the boundary is marked by a vertical shear zone (12 m wide) characterized by the development of abundant white mica and chlorite, folded by late, minor, brittle monoclinal and conjugate folds. Along the centre of the shear zone is a largely undeformed, concordant, virtually unfoliated 4 m wide pink granite sheet. The location of the boundary may well be controlled by the presence of an extensive marble calc-silicate horizon in the Strathy complex between the shear zone and a massive finely crystalline amphibolite sheet (fig. 1). On neither side of the shear zone is there any development of a 'platy fabric' such as is commonly found associated with basement sheets further west in Sutherland, nor have any such fabrics been found elsewhere in or near the complex. In particular it should be noted that where basic orthogneisses and amphibolites occur within the complex they appear to be integral to the complex as a whole, there are no signs of them having been tectonically emplaced into the grey gneisses.

## Field relations and petrography

The siliceous grey gneiss lithologies. These quartzrich rocks are all variations of the dominant grey gneiss lithology of quartz, sodic plagioclase, biotite ±garnet, and amphibole with accessory magnetite, apatite, epidote, and sphene. They are characterized by the absence of K-feldspar and are granoblastic, usually weakly banded or foliated. The plagioclase is frequently a sodic oligoclase, ranging to andesine in varieties containing hornblende. Garnet and amphiboles, both hornblendic and Ca-poor varieties are common in small quantities.

Quartz-garnet 'gneiss'. Where garnet is an important constituent, the lithology is a distinctive, white-weathering, virtually unfoliated quartz-garnet rock, usually with only minor amounts of biotite. This distinctive lithology is well displayed north of Armadale (fig. 1).

Orthoamphibole gneiss. In addition to hornblendic types, accessory amphiboles in the grey

gneisses include both monoclinic and orthorhombic Ca-poor types (cummingtonite-grunerite, gedrite-anthophyllite, as originally noted by Collins, 1941). Further optical work, and an XRD study by Dr A. G. Fraser of Hull University, have redefined the orthopyroxene reported by Harrison and Moorhouse (1976) as orthoamphibole. Garnetorthoamphibole gneiss is well displayed near Strathy lighthouse as a large lens or boudin, with a conspicuous rotated internal 'Z' schistosity, within normal grey gneiss. In this lithology garnet is abundant and biotite, which progressively replaces the amphibole, is more common towards the margins of the lens. In addition to a quartzoligoclase granoblastic matrix, garnet, orthamphibole, and staurolite appear to form an early paragenesis whilst biotite, sillimanite, and large blebby quartz postdates this. The early paragenesis is believed to represent the state of much of the complex after the first high-grade metamorphic episode (see later).

Cordierite gneiss. Only one occurrence of cordierite-bearing gneiss is known, this was discovered by Collins (1941) on the northern tip of Strathy Point. It is interesting that the only grey gneisses to contain K-feldspar also occur in this area, possibly due to metasomatism, which may also have produced the cordierite, 'pinite' and sillimanite which occur in narrow zones in a quartz-plagioclase-staurolite gneiss.

Magnetite gneiss. Magnetite, which is a common accessory in all the siliceous lithologies, forms conspicuous large euhedral crystals with a surrounding mafic-free halo in the quartz-magnetite gneiss, best seen north-east of Aultivullin (NC 814688, fig. 1). This was originally mapped by the Geological Survey as 'quartz-magnetite schist of Strathy' (Horne and Greenly, 1896).

'Granites'. All the above lithologies are cross-cut by large and small bodies of coarse-grained, weakly foliated sodic trondhjemite and by later, usually EW-trending, unfoliated pink microgranite sheets.

Mafic lithologies, hornblende gneiss. These are subordinate components of the complex and petrographically are grey gneisses with less quartz and more hornblende. Plagioclase tends to be andesine, reflecting the more calcic nature of the rocks. Often associated with the more massive amphibolite sheets, they were collectively termed 'Durcha-type Moine' by Collins (1941).

Amphibolites occur as large, apparently concordant sheets of hard, massive, quite fine-grained, hornblende-plagioclase-quartz rock frequently containing accessory clinopyroxene, e.g. at Port Mor (fig. 1). A spectacular net-veining of amphibolite by trondhjemite occurs on the east side of Armadale Bay (fig. 1). Possibly younger smaller

sheets of amphibolite, more or less schistose, still with some cross-cutting relationships to the grey gneiss, occur near A'Ghualann (NC 832668) on the west coast of Strathy Bay.

Armadale ultramafic body. This small body consists almost entirely of hornblende and opaque ore. It is distinct mineralogically from amphibolites and ultramafites found in the Sutherland Moine and in terms of its Ni and Cr values (Table VII), it is similar to pyroxene-rich ultramafics from the foreland Lewisian and analogous bodies from the orthogneissic basement sheets in central Sutherland (Moorhouse and Moorhouse, 1979a).

Marble and calc-silicate lithologies are best developed at Port Mor (fig. 1) and on Strathy Point. They vary from calcite-rich to scapolite-diopside calc-silicates; at Port Mor both types occur interbanded. The most characteristic paragenesis is scapolite, diopside, ±calcite, ±orange spinel with prominent accessory sphene, apatite, ore, and quartz, occasionally with acicular actinolite rimming the large pyroxenes.

Rock types in the Moine adjacent to the Strathy complex. The most striking feature on traversing from the complex into the adjacent rocks is the immediate incoming of abundant K-feldspar. To the west, the adjacent Kirtomy migmatitic biotite gneisses are well exposed and show a typical assemblage of varying proportions of oligoclase-andesine, K-feldspar, biotite, quartz, with some muscovite ± accessory garnet, sillimanite, chlorite, zircon, apatite, and ore. Amphiboles are never found in these Moine biotite gneisses.

To the east, the adjacent Moine rocks are not well exposed, being variably obscured by the Old Red Sandstone cover near the coast and inland by extensive peat bog. However, lithologies similar to Kirtomy biotite gneiss occur in the vicinity of Bowside Lodge (NC 830610) and in Strath Halladale (NC 896577) whereas around Portskerra (NC 874644) and the northern end of Strath Halladale the rocks are more siliceous, reminiscent of those to the west around Bettyhill.

All these Moine gneisses are distinct from the Strathy complex rocks in that they are more commonly strongly foliated and contain abundant K-feldspar and bodies of granite (s.s.), but no amphiboles, although sheets of meta-igneous amphibolite do occur (Moorhouse and Moorhouse, 1979a).

## **GEOCHEMISTRY**

Geochemical comparison of the Strathy complex with other gneiss groups

From the tables of analyses (Tables IV-VIII; for analytical technique see Moorhouse and Moor-

Table I. Mean analyses of the Strathy complex síliceous gneiss and other gneiss terrains.

|                                | A     | В     | C     | D     | E     | F     |
|--------------------------------|-------|-------|-------|-------|-------|-------|
| SiO2                           | 72.80 | 67.36 | 63.80 | 68.35 | 71.64 | 72.60 |
| TiO2                           | 0.25  | 0.64  | 0.31  | 0.53  | 0.82  | 0.68  |
| A1203                          | 12.46 | 14.61 | 15.09 | 13.83 | 11.80 | 11.8  |
| Fe <sub>2</sub> 0 <sub>3</sub> | 3.46  | 6.12  | 6.62  | 6.01  | 5.04  | 4.16  |
| MnO                            | 0.03  | 0.06  | 0.07  | 0.09  | 0.03  | 0.01  |
| Mg0                            | 1.62  | 2.86  | 2.81  | 1.90  | 2.04  | 1.54  |
| Ca0                            | 1.74  | 3.20  | 4.67  | 3.64  | 1.69  | 1.34  |
| Na <sub>2</sub> 0              | 5.27  | 2.52  | 3.09  | 4.67  | 2.90  | 2.79  |
| к <sub>2</sub> о               | 1.09  | 2.75  | 1.94  | 0.47  | 2.64  | 3.59  |
| P2O5                           | 0.07  | 0.13  | 0.15  | 0.09  | 0.16  | 0.14  |
| s                              | 824   | 800   | 1120  | 359   | 516   | 482   |
| Ga                             | 7     | -     | -     | 20    | 12    | 12    |
| Cr                             | 28    | 120   | 88    | 31    | 60    | 54    |
| Ni                             | 7     | 36    | 40    | 20    | 41    | 29    |
| Cu                             | 24    | -     | -     | 40    | 13    | 12    |
| Zn                             | 62    | -     | -     | 79    | 77    | 55    |
| RЪ                             | 20    | 59    | 42    | 9     | 91    | 113   |
| Sr                             | 200   | 405   | 424   | 149   | 365   | 353   |
| Y                              | 26    | 19    | 13    | -     | 31    | 28    |
| Zr                             | 140   | 288   | 218   | 133   | 430   | 455   |
| NЪ                             | 4     | 11    | 7     | -     | 14    | 14    |
| Ва                             | 273   | 1100  | 802   | 206   | 758   | 828   |
| La                             | 7     | -     | 31    | -     | 40    | 37    |
| Ce                             | 37    | 51    | 56    | -     | 108   | 107   |
| Pb                             | 11    | 20    | 16    | 19    | 21    | 27    |
| Th                             | 3     | 7     | 7     | 8     | 16    | 18    |
| K/Rb                           | 452   | 387   | 383   | 433   | 241   | 264   |
| K/Rb*                          | 528   | -     | -     | -     | 247   | 283   |
| Rb/Sr                          | 0.100 | 0.146 | 0.099 | 0.060 | 0.249 | 0.320 |
| Rb/Sr*                         | 0.123 | -     | -     | -     | 0.344 | 0.380 |
| Ca/Y                           | 479   | 1204  | 2569  | -     | 390   | 342   |
| Ca/Y*                          | 626   | -     | -     | -     | 551   | 409   |

 $\operatorname{Fe}_2 \operatorname{O}_3$  is total Fe as  $\operatorname{Fe}_2 \operatorname{O}_3$ . K/Rb is mean K/mean Rb, K/Rb\* is mean K/Rb.

- A. Strathy complex siliceous gneisses, 50 analyses, Table IV.
- B. Leverburgh belt metasediments, 21 analyses, Sheraton et al. (1973).
- C. East Greenland metasediments, 48 analyses, Sheraton et al. (1973).
- D. Acid-intermediate gneisses, Tromøy, 79 analyses, Cooper and Field
- E. Kirtomy assemblage biotite gneisses, 39 analyses, Moorhouse and Moorhouse (unpublished data).
- F. Bettyhill assemblage quartzitic gneisses, 33 analyses, Moorhouse and Moorhouse (unpublished data).

house, 1979a) and a comparison of the mean analysis of the Strathy siliceous gneisses with averages from other terrains (Table I), it is evident that these Strathy rocks are SiO<sub>2</sub>-rich; concomitantly TiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, total Fe<sub>2</sub>O<sub>3</sub>, MnO, CaO tend to be lower, but Na<sub>2</sub>O is higher than other high-grade metamorphic rock groups. Potassium is even lower than the average for the Scourian rocks of Assynt but not as low as the Tromøy acid-intermediate gneiss (Tables I-II). In Table II average analyses of various granulite terrains are compared with a weighted average of Strathy grey gneiss, amphibolite, and hornblende gneiss, representing a crude mean of the 'non-granitic' part of the complex.

Several interesting points emerge from this comparison of the mean analyses which are probably

Table II. Mean analyses of various granulite terrains, weighted averages for the Strathy complex, and the mean analysis of the Strathy trondhjemites.

|                                | A     | B     | ε     | α     | E     | F     | G     |
|--------------------------------|-------|-------|-------|-------|-------|-------|-------|
| SiO <sub>2</sub>               | 64.56 | 69.35 | 65.58 | 60.70 | 60.60 | 71.54 | 69.66 |
| TiO2                           | 0.47  | 0.37  | 0.65  | 0.95  | 0.90  | 0.24  | 0.23  |
| A1203                          | 15.75 | 12.89 | 14.18 | 15.83 | 15.40 | 13.79 | 15.79 |
| Fe <sub>2</sub> O <sub>3</sub> | 4.69  | 4.67  | 7.11  | 7.82  | 7.20  | 2.89  | 1.98  |
| Mn0                            | 0.06  | 0.04  | 0.11  | 0.11  | 0.20  | 0.02  | 0.00  |
| Mg0                            | 2.23  | 2.55  | 2.50  | 3.02  | 3.90  | 1.54  | 1.42  |
| CaO                            | 4.50  | 2.66  | 4.16  | 5.28  | 5.70  | 1.89  | 2.13  |
| Na <sub>2</sub> O              | 4.60  | 5.04  | 4.59  | 3.55  | 2.80  | 5.63  | 6.16  |
| K <sub>2</sub> 0               | 1.15  | 1.08  | 0.47  | 2.28  | 2.60  | 1.23  | 1.45  |
| P205                           | 0.16  | 0.09  | 0.15  |       | 0.20  | 0.10  | 0.14  |
| s                              | -     | 826   | 514   | -     | -     | 736   | 604   |
| Ga                             | -     | 9     | 21    | -     | -     | 9     | 13    |
| Cr                             | 48    | 62    | 29    | 71    | -     | 30    | 34    |
| Ni                             | 37    | 29    | 19    | -     | -     | 8     | 10    |
| Cu                             | -     | 26    | 46    | -     | -     | 17    | 7     |
| Zn                             | -     | 70    | 94    | -     | -     | 54    | 43    |
| Rb                             | 13    | 19    | 8     | 46    | 70    | 23    | 28    |
| Sr                             | 565   | 221   | 160   | 543   | 340   | 442   | 804   |
| Y                              | 8     | 25    | -     | -     | -     | 17    | 4     |
| Zr                             | 197   | 146   | 123   | 222   | 310   | 173   | 222   |
| NЬ                             | 5     | 4     | -     | -     | -     | 4     | 4     |
| Ва                             | 779   | 262   | 290   | 1217  | 1090  | 438   | 685   |
| La                             | -     | 8     | -     | -     | -     | 10    | 14    |
| Ce                             | 42    | 42    | -     | -     | -     | 39    | 41    |
| РЪ                             | 12    | 11    | 19    | -     | 20    | 16    | 23    |
| Th                             | 1     | 3     | 7     | -     | 2     | 4     | 5     |
| K/Rb                           | 734   | 471   | 487   | 411   | 308   | 444   | 430   |
| K/Ba                           | 12.25 | 34.21 | 18.58 | 15.56 | 19.80 | 23.30 | 17.57 |
| Rb/Sr                          | 0.023 | 0.086 | 0.050 | 0.080 | 0.206 | 0.052 | 0.035 |
| K/Sr                           | 16.90 | 40.57 | 24.38 | 34.86 | 63.48 | 23.10 | 14.97 |
| Ca/Y                           | 4022  | 761   | -     | -     | -     | 795   | 3772  |

- A. Scourian gneisses from Assynt, 154 analyses, Sheraton et al. (1973).
- B. Strathy complex weighted average, 85% siliceous gneiss, 12% amphibolite, 3% hornblende gneiss.
- C. Trommy granulites, 95 analyses, Cooper and Field (1977).
- D. Brazilian granulites, 62 analyses, Sighinolfi (1971).
- E. Musgrave Range granulites, 23 analyses, Lambert and Heier (1968).
- F. Strathy complex, weighted average, 60% siliceous gneiss,  $40\,\%$  troudhjemite.
- G. Trondhjemites in Strathy complex, 22 analyses, Moorhouse and Moorhouse (in preparation).

significant, in spite of the obvious inherent disadvantages of such comparisons. First, the Strathy complex as a whole, with or without the trondhiemites, is more siliceous than the other highgrade terrains. Secondly, the complex is low in K<sub>2</sub>O, Rb, and Ba, especially compared with the Brazilian and Musgrave granulites. The Strathy siliceous gneisses are also quite different from the average Leverburgh belt and East Greenland metasediments (Table I). In particular K<sub>2</sub>O, t·Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, MgO, CaO and P<sub>2</sub>O<sub>5</sub>, Cr, Ni, Rb, Sr, Zr, Ba, Ce, Pb, and Th are considerably lower, both in absolute values and relative to SiO2 content, whereas SiO<sub>2</sub>, Na<sub>2</sub>O and (mean K/mean Rb) are higher. The Tromøy acid-intermediate gneisses are more akin to the Strathy siliceous gneisses in that Ni, Cr, Sr, Zr, and Ba are similar, SiO<sub>2</sub> and Na<sub>2</sub>O

| Table III. | Mean analyses of basic hornblendic rocks from the |
|------------|---|
|            | Strathy complex and the Bettyhill and Kirtomy     |
|            | Moine assemblages.                                |

|                                | A          | В     | С     |
|--------------------------------|------------|-------|-------|
| sio <sub>2</sub>               | 51.48      | 49.40 | 49.31 |
| TiO <sub>2</sub>               | 1.04       | 1.02  | 2.03  |
| A1203                          | 15.33      | 15.33 | 12.94 |
| Fe <sub>2</sub> O <sub>3</sub> | 9.81       | 11.92 | 14.55 |
| MnO                            | 0.12       | 0.12  | 0.05  |
| Mg0                            | 6.70       | 8.10  | 6.57  |
| CaO                            | 8.23       | 7.82  | 8.84  |
| Na <sub>2</sub> 0              | 4.18       | 3.67  | 2.10  |
| K <sub>2</sub> 0               | 1.03       | 1.00  | 1.40  |
| P2O5                           | 0.19       | 0.24  | 0.25  |
| S                              | 944        | 809   | 1063  |
| Ga                             | 19         | 20    | 26    |
| Cr                             | 159        | 276   | 193   |
| Ni                             | 102        | 170   | 144   |
| Cu                             | <b>5</b> 5 | 30    | 43    |
| Zn                             | 101        | 123   | 131   |
| Rb                             | 15         | 13    | 42    |
| Sr                             | 283        | 355   | 176   |
| Y                              | 22         | 20    | 51    |
| Zr                             | 152        | 191   | 232   |
| ΝЪ                             | 6          | 7     | 10    |
| Ва                             | 218        | 197   | 340   |
| La                             | 9          | 11    | . 9   |
| Се                             | 62         | 70    | 75    |
| Рb                             | 8          | 8     | 10    |
| Th                             | 3          | 1     | 4     |

- A. Strathy complex hornblende gneiss, 13 analyses, Table VI.
- B. Strathy complex amphibolites, 13 analyses, Table VII.
- C. Amphibolites from the Bettyhill and Kirtomy Moine assemblages, 13 analyses, Moorhouse and Moorhouse (1979; Table VI, BS277- BS195 Bettyhill, BS 69-BS100 Kirtomy).

are almost as high, but  $t \cdot Fe_2O_3$  and CaO are much higher, whilst  $K_2O$  and Rb are considerably lower (compare columns A and D Table I, columns B and C Table II).

# Geochemical comparison of the Strathy complex with the local Moine rocks

The Strathy complex siliceous gneisses have an  $SiO_2$  content equivalent to the local Moine gneisses but have higher average  $Na_2O$ , lower  $TiO_2$ ,  $t \cdot Fe_2O_3$ ,  $K_2O$ , Cr, Ni, Sr, Zr, Nb, Pb, and Th, notably lower Rb, Ba, La, and Ce, whilst the Strathy K/Rb ratios are considerably higher and Rb/Sr much lower (Table I).

The SiO<sub>2</sub>-TiO<sub>2</sub> and t·Fe<sub>2</sub>O<sub>3</sub>-TiO<sub>2</sub> plots (fig. 2) illustrate some of these differences. On the SiO<sub>2</sub>-TiO<sub>2</sub> plot the Strathy gneisses occupy a low-TiO<sub>2</sub> field comparable with that of Lewisian lithologies from basement sheets within the Moine further

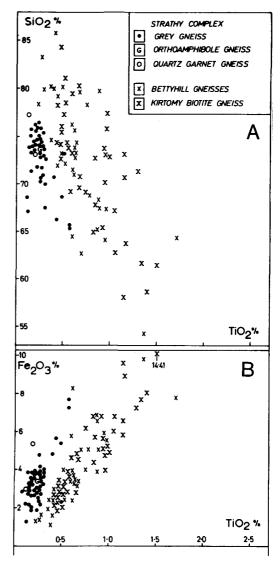


Fig. 2. (A) SiO<sub>2</sub> vs. TiO<sub>2</sub>. (B) Total Fe<sub>2</sub>O<sub>3</sub> vs. TiO<sub>2</sub>.

west in Sutherland (Moorhouse, 1976; Moorhouse and Moorhouse, 1977, and in prep.). The Moine gneisses exhibit a much greater range of  $SiO_2$ ,  $t \cdot Fe_2O_3$  and  $TiO_2$  values with a more pronounced negative  $SiO_2$ - $TiO_2$  correlation and positive  $t \cdot Fe_2O_3$ - $TiO_2$  correlation than those of the Strathy gneisses.

The K<sub>2</sub>O vs. Na<sub>2</sub>O diagram (fig. 3A) emphasizes the low-K<sub>2</sub>O, high-Na<sub>2</sub>O nature of the Strathy gneisses which is quite distinct from the lower-Na<sub>2</sub>O, higher-K<sub>2</sub>O of the Moine gneisses. In a similar manner the Moine rocks occupy a separate high Rb and Sr field on fig. 3B.

Table IV. The Strathy complex siliceous gneisses.

|                                | BS 31  | BS 26  | BS 43  | BS131  | BS 50  | BS 4   | BS 63  | BS117  | BS128  | BS137  | BS 82  | B\$121 | BS 48  | BS127  |                   |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|-------------------|
| SiO,                           | 78.10  | 77.26  | 76.47  | 76.47  | 76.26  | 76.10  | 75.86  | 75.86  | 75.70  | 75.66  | 75.37  | 75.18  | 75.06  | 74.97  | sio <sub>2</sub>  |
| TiO,                           | 0.14   | 0.13   | 0.24   | 0.24   | 0.19   | 0.21   | 0.22   | 0.26   | 0.16   | 0.21   | 0.29   | 0.18   | 0.15   | 0.24   | TiO2              |
| A1203                          | 10.20  | 9.89   | 10.94  | 11.75  | 11.65  | 10.63  | 11.33  | 11.46  | 12.03  | 12.02  | 11.47  | 12.12  | 11.49  | 12.92  | A1203             |
| Fe <sub>2</sub> 0 <sub>3</sub> | 2.16   | 2.92   | 2.66   | 2.85   | 2.66   | 3.79   | 3.72   | 2.35   | 2.74   | 2.65   | 3.30   | 1.98   | 2.22   | 2.97   | Fe203             |
| MnO                            | 0.05   | 0.05   | 0.04   | 0.00   | 0.04   | 0.00   | 0.03   | 0.00   | 0.00   | 0.00   | 0.00   | 0.08   | 0.00   | 0.03   | MnO               |
| MgO                            | 1.30   | 2.08   | 1.39   | 1.00   | 1.50   | 1.76   | 1.59   | 1.10   | 1.05   | 0.78   | 0.88   | 1.30   | 1.26   | 1.33   | MgO               |
| CaO                            | 0.90   | 0.58   | 0.98   | 0.98   | 2.42   | 0.98   | 1.36   | 1.35   | 2.42   | 1.51   | 1.59   | 1.24   | 1.83   | 1.78   | Ca0               |
| Na <sub>2</sub> 0              | 5.80   | 5.40   | 5.10   | 5.70   | 5.00   | 4.70   | 5.00   | 4.60   | 4.60   | 5.40   | 5.40   | 5.90   | 4.50   | 5.80   | Na <sub>2</sub> 0 |
| к,0                            | 0.51   | 0.56   | 1.22   | 1.10   | 1.01   | 1.02   | 1.20   | 1.46   | 0.55   | 0.91   | 0.72   | 0.77   | 1.75   | 0.98   | к20               |
| P205                           | 0.02   | 0.01   | 0.02   | 0.09   | 0.03   | 0.05   | 0.04   | 0.09   | 0.04   | 0.08   | 0.07   | 0.03   | 0.06   | 0.09   | P2O5              |
| Loss                           | 0.75   | 0.86   | 0.80   | 1.72   | 0.94   | 0.73   | 1.10   | 1.51   | 0.58   | 0.69   | 0.91   | 1.02   | 1.45   | 0.75   | Loss              |
| Total                          | 99.93  | 99.74  | 99.86  | 101.90 | 101.70 | 99.97  | 101.45 | 100.04 | 99.87  | 99.91  | 100.00 | 99.80  | 99.77  | 101.86 | Total             |
| s                              | 410    | 910    | 630    | 290    | 530    | 820    | 30     | 440    | 570    | 350    | 280    | 1070   | 1060   | 210    | s                 |
| Ga                             | 4      | 4      | 0      | 0      | 6      | 6      | 7      | 1      | 0      | 6      | 5      | 2      | 3      | 8      | Ga                |
| Cr                             | 33     | 27     | 28     | 26     | 29     | 29     | 30     | 33     | 30     | 26     | 29     | 27     | 31     | 32     | Cr                |
| Ni                             | 15     | 1      | 3      | 4      | 5      | 8      | 1      | 4      | 4      | 3      | 9      | 4      | 0      | 8      | Ni                |
| Cu                             | 58     | 18     | 7      | 0      | 21     | 4      | 6      | 15     | 0      | 48     | 15     | 58     | 7      | 3      | Cu                |
| Zn                             | 28     | 10     | 57     | 0      | 52     | 36     | 68     | 85     | 0      | 58     | 52     | 60     | 51     | 34     | Zn                |
| RЪ                             | 5      | . 7    | 13     | 23     | 12     | 18     | 30     | 32     | 9      | 15     | 11     | 6      | 25     | 16     | Rb                |
| Sr                             | 87     | 56     | 251    | 111    | 266    | 230    | 232    | 180    | 188    | 263    | 325    | 124    | 84     | 264    | Sr                |
| Y                              | 28     | 27     | 14     | 31     | 30     | 34     | 6      | 31     | 31     | 27     | 21     | 28     | 24     | 26     | Y                 |
| Zr                             | 114    | 108    | 201    | 111    | 123    | 156    | 134    | 172    | 117    | 118    | 122    | 118    | 87     | 132    | Zr                |
| Nb                             | 3      | 4      | 2      | 1      | 1      | 1      | 6      | 3      | 2      | 3      | 4      | 3      | 4      | 2      | Nb                |
| Ba                             | 142    | 187    | 293    | 240    | 219    | 284    | 396    | 382    | 178    | 296    | 244    | 204    | 220    | 284    | Ba                |
| La                             | 7      | 8      | 2      | 7      | 7      | 7      | 6      | 10     | 7      | 9      | 4      | 7      | 7      | 7      | La                |
| Ce                             | 34     | 40     | 23     | 28     | 33     | 30     | 31     | 39     | 39     | 30     | 25     | 33     | 35     | 39     | Ce                |
| Pb                             | 9      | 8      | 12     | 8      | 9      | 8      | 12     | 14     | 9      | 11     | 12     | 6      | 8      | 7      | Pb                |
| Th                             | 2      | 2      | 0      | 1      | 2      | 0      | 3      | 3      | 4      | 2      | 2      | 2      | 4      | 3      | Th                |
| NC                             | 804661 | 801656 | 829695 | 818682 | 876694 | 785655 | 812685 | 831675 | 813671 | 812687 | 804643 | 832671 | 828697 | 816662 | NC                |

NC Initial letters of National Grid Reference.

 ${\rm Fe_2O_3}$  is total Fe as  ${\rm Fe_2O_3}$ . Loss is loss on ignition of dried samples. BS 26 is a quartz-garnet gneiss.

|                                | BS122  | BS130  | BS 51  | BS 37  | BS191  | B\$ 28 | BS125  | BS 39  | BS 40  | BS 97  | BS 96  | BS 22  | BS 64  | BS 3   |                                |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------------------------------|
| SiO,                           | 74.64  | 74.63  | 74.52  | 74.48  | 74.45  | 74.05  | 73.92  | 73.90  | 73.80  | 73.80  | 73.68  | 73.62  | 73.36  | 73.35  | SiO <sub>2</sub>               |
| TiO,                           | 0.25   | 0.22   | 0.23   | 0.26   | 0.19   | 0.19   | 0.28   | 0.18   | 0.16   | 0.28   | 0.22   | 0.25   | 0.19   | 0.19   | TiO2                           |
| A1203                          | 13.10  | 12.05  | 12.16  | 12.72  | 11.58  | 11.78  | 11.92  | 12.79  | 12.09  | 12.84  | 12.65  | 11.81  | 12.41  | 12.05  | A1203                          |
| Fe <sub>2</sub> 0 <sub>3</sub> | 2.46   | 2.93   | 3.75   | 2.90   | 3.38   | 3.33   | 3.10   | 1.91   | 3.01   | 3.41   | 3.78   | 3.58   | 3.38   | 5.36   | Fe <sub>2</sub> 0 <sub>3</sub> |
| Mn0                            | 0.00   | 0.08   | 0.06   | 0.00   | 0.00   | 0.08   | 0.01   | 0.00   | 0.00   | 0.07   | 0.00   | 0.00   | 0.15   | 0.08   | Mn0                            |
| Mg0                            | 2.04   | 0.79   | 2.05   | 1.03   | 1.16   | 1.34   | 1.78   | 1.02   | 1.15   | 2.14   | 1.37   | 2.64   | 1.49   | 1.16   | MgO                            |
| Ca0                            | 1.31   | 0.96   | 1.40   | 1.82   | 1.44   | 2.03   | 0.94   | 1.48   | 1.91   | 1.72   | 0.81   | 3.17   | 1.05   | 0.72   | CaO                            |
| Na <sub>2</sub> O              | 5.90   | 5.80   | 5.40   | 5.30   | 6.00   | 4.20   | 5.40   | 6.30   | 5.30   | 4.40   | 6.10   | 3.10   | 6.00   | 6.10   | Na <sub>2</sub> 0              |
| K <sub>2</sub> 0               | 1.34   | 1.13   | 0.74   | 1.22   | 0.70   | 1.19   | 1.58   | 0.86   | 1.02   | 1.83   | 1.30   | 0.53   | 0.78   | 0.21   | K <sub>2</sub> O               |
| P205                           | 0.04   | 0.03   | 0.05   | 0.06   | 0.04   | 0.03   | 0.05   | 0.03   | 0.05   | 0.11   | 0.08   | 0.08   | 0.03   | 0.06   | P205                           |
| Loss                           | 0.33   | 0.84   | 0.43   | 0.84   | 0.49   | 1.74   | 0.58   | 1.58   | 1.31   | 1.24   | 1.47   | 1.02   | 1.18   | 0.40   | Loss                           |
| Total                          | 101.41 | 99.46  | 100.79 | 100.66 | 99.43  | 99.96  | 99.56  | 100.05 | 99.80  | 101.84 | 101.46 | 99.80  | 100.02 | 101.68 | Total                          |
| s                              | 550    | 600    | 290    | 670    | 560    | 140    | 620    | 400    | 1100   | 470    | 1020   | 1040   | 110    | 310    | s                              |
| Ga                             | 0      | 7      | 8      | 8      | 5      | 6      | 8      | 8      | 5      | 7      | 16     | 11     | 2      | 6      | Ga                             |
| Cr                             | 35     | 28     | 29     | 29     | 26     | 29     | 30     | 38     | 29     | 46     | 31     | 26     | 27     | 22     | Cr                             |
| Ni                             | 8      | 24     | 2      | 25     | 5      | 3      | 6      | 9      | 3      | 16     | 4      | 2      | 3      | 3      | Ni                             |
| Cu                             | 0      | 14     | 11     | 14     | 12     | 12     | 18     | 10     | 9      | 5      | 6      | 3      | 43     | 21     | Cu                             |
| Zn                             | 0      | 64     | 62     | 57     | 28     | 52     | 84     | 33     | 57     | 116    | 84     | 80     | 274    | 84     | Zn                             |
| RЬ                             | 28     | 25     | 8      | 30     | 9      | 22     | 19     | 16     | 22     | 43     | 32     | 5      | 13     | 1      | Rb                             |
| Sr                             | 213    | 158    | 84     | 423    | 171    | 298    | 135    | 316    | 244    | 286    | 71     | 216    | 121    | 87     | Sr                             |
| Y                              | 8      | 31     | 28     | 5      | 33     | 26     | 25     | 4      | 25     | 25     | 41     | 29     | 45     | 36     | ¥                              |
| Zr                             | 184    | 150    | 113    | 129    | 125    | 107    | 202    | 151    | 151    | 116    | 171    | 102    | 130    | 116    | Zr                             |
| Nb                             | 6      | 0      | 3      | 4      | 1      | 3      | 6      | 4      | 1      | 3      | 8      | 2      | 1      | 1      | Nb                             |
| Ва                             | 423    | 392    | 155    | 513    | 267    | 193    | 252    | 431    | 339    | 242    | 238    | 134    | 180    | 131    | Ba                             |
| La                             | 5      | 6      | 6      | 11     | 8      | 5      | 10     | 13     | 8      | 7      | 7      | 9      | 11     | 10     | La                             |
| Ce                             | 33     | 29     | 38     | 52     | 37     | 33     | 46     | 38     | 31     | 27     | 48     | 38     | 37     | 44     | Ce                             |
| Рb                             | 18     | 11     | 8      | 16     | 6      | 7      | 12     | 17     | 10     | 9      | 14     | 7      | 5      | . 6    | Pb                             |
| Th                             | 4      | 1      | 2      | 4      | 2      | 2      | 5      | 8      | 2      | 2      | 5      | . 2    | 1      | 2      | Th                             |
| NC                             | 831669 | 821663 | 831687 | 822649 | 775655 | 799653 | 833668 | 813686 | 812687 | 831683 | 830683 | 828695 | 812687 | 787653 | NC                             |

BS 22 is an orthoamphibole gneiss. BS 3 is a quartz-garnet gneiss.

Table IV (continued). The Strathy complex siliceous gneisses.

| -                              | BS 42  | BS 65  | BS115  | BS221  | BS 58  | BS 46  | BS 35  | BS 66  | BS120  | BS 61  | BS 45  | BS222  | BS 41  | BS133  |                                |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------------------------------|
| sio <sub>2</sub>               | 73.34  | 73.25  | 73.18  | 72.87  | 72.81  | 72.70  | 71.76  | 71.68  | 71.28  | 70.89  | 70.70  | 70.61  | 70.29  | 69.92  | SiO <sub>2</sub>               |
| TiO2                           | 0.24   | 0.16   | 0.50   | 0.28   | 0.27   | 0.31   | 0.22   | 0.27   | 0.16   | 0.28   | 0.40   | 0.27   | 0.26   | 0.31   | TiO2                           |
| A1203                          | 12.78  | 12.38  | 12.79  | 135    | 12.34  | 12.10  | 11.75  | 12.45  | 13.19  | 12.34  | 11.28  | 14.45  | 12.93  | 13.20  | A1 <sub>2</sub> 0 <sub>3</sub> |
| Fe <sub>2</sub> O <sub>3</sub> | 2.74   | 3.16   | 2.31   | 2.72   | 4.14   | 3.62   | 3.73   | 3.82   | 3.00   | 3.56   | 4.80   | 3.78   | 4.75   | 3.98   | Fe <sub>2</sub> O <sub>3</sub> |
| MnO                            | 0.00   | 0.00   | 0.03   | 0,00   | 0.09   | 0.00   | 0.00   | 0.00   | 0.00   | 0.00   | 0.00   | 0.00   | 0.17   | 0.00   | MnO                            |
| MgO                            | 1.03   | 1.45   | 1.84   | 0.99   | 2.69   | 1.49   | 1.23   | 1.63   | 3.72   | 1.42   | 3.82   | 0.72   | 2.25   | 1.83   | MgO                            |
| CaO                            | 2.09   | 1.74   | 1.90   | 1.87   | 1.15   | 2.70   | 3.01   | 1.30   | 1.42   | 1.20   | 1.42   | 2.05   | 1.04   | 1.76   | CaO                            |
| Na <sub>2</sub> 0              | 4.90   | 5.20   | 4.60   | 5.70   | 5.70   | 4.40   | 4.20   | 6.10   | 3.80   | 6.20   | 3.60   | 6.40   | 5.50   | 5.60   | Na <sub>2</sub> 0              |
| к <sub>2</sub> о               | 1.26   | 0.97   | 2.13   | 0.59   | 1.00   | 1.35   | 0.97   | 1.14   | 2,28   | 0.96   | 2.30   | 0.92   | 1.64   | 1.36   | K20                            |
| P2O5                           | 0.06   | 0.04   | 0.03   | 0.11   | 0.05   | 0.08   | 0.05   | 0.05   | 0.03   | 0.10   | 0.02   | 0.08   | 0.06   | 0.13   | P205                           |
| Loss                           | 1.48   | 1.45   | 0.51   | 1.03   | 1.53   | 1.17   | 1.49   | 1.37   | 0.80   | 3.01   | 1.45   | 0.81   | 1.05   | 1.30   | Loss                           |
| Total                          | 99.92  | 99.80  | 99.82  | 99.31  | 101.77 | 99.92  | 98.41  | 99.81  | 99.68  | 99.96  | 99.79  | 100.09 | 99.94  | 99.39  | Total                          |
| s                              | 850    | 1010   | 960    | 200    | 170    | 940    | 4840   | 990    | 450    | 840    | 1310   | 140    | 370    | 240    | s                              |
| Ga                             | 8      | 5      | 0      | 8      | 6      | 11     | 9      | 9      | 9      | 12     | 11     | 10     | 9      | 12     | Ga                             |
| Cr                             | 35     | 28     | 27     | 28     | 33     | 28     | 29     | 30     | 27     | 27     | 30     | 25     | 25     | 27     | Cr                             |
| Ni                             | 13     | 2      | 5      | 7      | 4      | 4      | 9      | 2      | 5      | 4      | 2      | 5      | 2      | 4      | Ni                             |
| Cu                             | 18     | 3      | 40     | 58     | 6      | 36     | 182    | 10     | 2      | 18     | 124    | 8      | 17     | 7      | Cu                             |
| Zn                             | 51     | 55     | 144    | 44     | 54     | 70     | 20     | 60     | 74     | 41     | 121    | 56     | 56     | 61     | Zn                             |
| Rb                             | 18     | 16     | 34     | 11     | 13     | 23     | 27     | 23     | 30     | 21     | 38     | 22     | 27     | 32     | Rb                             |
| Sr                             | 228    | 191    | 234    | 128    | 63     | 314    | 194    | 186    | 136    | 82     | 195    | 178    | 66     | 129    | Sr                             |
| Y                              | 26     | 30     | 26     | 18     | 34     | 25     | 27     | 19     | 41     | 33     | 22     | 34     | 34     | 29     | Υ.                             |
| Zr                             | 106    | 93     | 232    | 127    | 104    | 122    | 100    | 80     | 163    | 108    | 157    | 127    | 126    | 107    | Zr                             |
| Nb                             | 0      | 0      | 5      | 3      | 3      | 3      | 3      | 3      | 4      | 2      | 2      | 5      | 3      | 0      | Nb                             |
| Ba                             | 348    | 232    | 543    | 141    | 161    | 314    | 251    | 280    | 163    | 233    | 390    | 277    | 177    | 216    | Ва                             |
| La                             | 8      | 5      | 9      | 8      | 7      | 14     | 8      | 5      | 8      | 6      | 4      | 9      | 5      | 5      | La                             |
| Ce                             | 41     | 31     | 25     | 42     | 41     | 68     | 34     | 25     | 42     | 32     | 23     | 47     | 31     | 38     | Ce                             |
| Pb                             | 14     | 10     | 35     | 9      | 5      | 10     | 8      | 9      | 10     | 11     | 11     | 9      | 6      | 7      | Pъ                             |
| Th                             | 4      | 4      | 2      | 4      | 1      | 8      | 4      | 0      | 3      | 1      | 2      | 2      | 3      | 1      | Th                             |
| NC                             | 829698 | 818687 | 831675 | 802660 | 823689 | 827698 | 806664 | 820671 | 831675 | 821688 | 828699 | 823650 | 815687 | 816688 | NC                             |

Table IV (continued). The Strathy complex siliceous gneisses.

| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   |          |
|---|----------|
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 29 BS13  |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | 41 65.30 |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 57 0.14  |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 27 15.72 |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 58 3.8   |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 18 0.0   |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 89 3.3   |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | 19 3.69  |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | 6.60     |
| Loss     0.88     8.26     4.75     1.31     3.50     0.61     0.       Total     100.00     99.85     99.87     99.96     99.10     100.78     99.       S     30     200     350     180     9999     620     10       Ga     13     5     9     11     9     17     17 | 75 0.33  |
| Total 100.00 99.85 99.87 99.96 99.10 100.78 99.<br>S 30 200 350 180 9999 620 10<br>Ga 13 5 9 11 9 17  | 25 0.03  |
| S 30 200 350 180 9999 620 Io<br>Ga 13 5 9 11 9 17   | 37 0.72  |
| Ga 13 5 9 11 9 17   | 96 99.80 |
|   | 00 940   |
| Cr 26 30 25 26 25 17  | 16 15    |
|   | 17 30    |
| Ni 28 17 4 5 10 0   | 0 16     |
| Cu 22 6 17 19 157 0   | 1 :      |
| Zn 43 51 87 60 53 67 1:   | 25 50    |
| Rb 9 16 30 35 18 17 .   | 54       |
| Sr 395 86 361 400 321 218 1   | 24       |
| Y 35 16 6 13 19 32  | 2 35     |
| Zr 100 629 120 146 84 94 1  | 24 179   |
| Nb 3 23 3 6 7 2   | 8 :      |
| Ba 253 256 380 511 307 235 3  | 90 120   |
| La 6 4 8 5 5 5  | 6 12     |
| Ce 30 22 48 26 39 47  | 48 50    |
| Pb 13 9 14 11 12 8 :  | 21 11    |
| rh 0 3 5 5 2 2  | 6 4      |
| NC 819657 812685 795640 803647 826692 793633 8226   |          |

Table V. Calc-silicates from the Bettyhill assemblage (BS224A, BS224B) and the Kirtomy assemblage (BS 88, BS 89).

|                                | BS224A | BS224B | BS 88  | BS 89  |
|--------------------------------|--------|--------|--------|--------|
| SiO <sub>2</sub>               | 69.64  | 77.25  | 64.67  | 67.41  |
| TiO2                           | 0.65   | 0.31   | 0.67   | 0.28   |
| A1,03                          | 11.40  | 9.66   | 15.73  | 13.96  |
| Fe <sub>2</sub> O <sub>3</sub> | 3.18   | 3.04   | 4.98   | 2.93   |
| MnO                            | 0.03   | 0.13   | 0.25   | 0.10   |
| MgO                            | 1.21   | 1.82   | 2.34   | 0.56   |
| CaO                            | 4.22   | 2.52   | 6.31   | 7.22   |
| Na <sub>2</sub> O              | 1.50   | 1.30   | 0.60   | 0.90   |
| K <sub>2</sub> 0               | 2.30   | 2.23   | 2.62   | 1.56   |
| P205                           | 0.17   | 0.18   | 0.23   | 0.30   |
| Loss                           | 5.52   | 1.72   | 1.62   | 4.67   |
| Total                          | 99.82  | 100.16 | 100.02 | 99.89  |
| s                              | 240    | 770    | 1150   | 270    |
| Ga                             | 15     | 7      | 22     | 34     |
| Cr                             | 53     | 40     | 53     | 33     |
| Ni                             | 26     | 21     | 27     | 10     |
| Cu                             | 5      | 11     | 5      | 0      |
| Zn                             | 34     | 28     | 97     | 15     |
| Rb                             | 98     | 109    | 82     | 54     |
| Sr                             | 415    | 260    | 314    | 280    |
| Y                              | 39     | 21     | 42     | 17     |
| Zr                             | 349    | 137    | 369    | 194    |
| NЪ                             | 17     | 10     | 17     | 9      |
| Ba                             | 787    | 631    | 199    | 232    |
| La                             | 44     | 38     | 0      | 17     |
| Ce                             | 122    | 124    | 73     | 61     |
| РЬ                             | 15     | 18     | 19     | 22     |
| Th                             | 13     | ,7     | 19     | 8      |
| NC                             | 710620 | 710620 | 773617 | 772613 |

Table VI. The Strathy complex hornblende gneisses.

|                                | BS123  | BS 21  | BS113  | BS293  | BS256A | BS 54  | BS 47  | BS 15  | BS 25  | BS 36  | BS119  | BS 81  | BS 53  |                   |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|-------------------|
| SiO2                           | 57.04  | 56.39  | 55.17  | 54.88  | 52.74  | 52.09  | 51.83  | 51.73  | 48.73  | 48.20  | 47.62  | 46.52  | 46.28  | sio <sub>2</sub>  |
| TiO <sub>2</sub>               | 0.88   | 0.38   | 1.57   | 1.39   | 0.60   | 0.42   | 0.49   | 0.94   | 0.65   | 2.00   | 1.50   | 0.65   | 1.99   | TiO <sub>2</sub>  |
| A1203                          | 15.25  | 15.99  | 17,18  | 14.00  | 16.24  | 14.89  | 14.29  | 15.19  | 15.57  | 13.32  | 17.18  | 14.51  | 15.72  | A1203             |
| Fe <sub>2</sub> O <sub>3</sub> | 8.93   | 6.39   | .8.41  | 11.06  | 9.25   | 10.21  | 9.35   | 10.23  | 7.64   | 13.61  | 9.61   | 11.67  | 11.22  | Fe203             |
| MnO                            | 0.19   | 0.23   | 0.09   | 0.00   | 0.01   | 0.01   | 0.14   | 0.16   | 0.17   | 0.12   | 0.11   | 0,13   | 0.14   | MnO               |
| MgO                            | 4.85   | 5.20   | 3.83   | -5.28  | 7.99   | 7.29   | 7.74   | 5.88   | 7.16   | 7.14   | 8.35   | 8,66   | 7.74   | Mg0               |
| CaO                            | 5.78   | 7.01   | 4.59   | 8.13   | 5.81   | 7.55   | 9.92   | 6.50   | 6.80   | 10,12  | 12.23  | 11.69  | 10.92  | CaO               |
| Na <sub>2</sub> 0              | 4.90   | 4.80   | 5.10   | 2.80   | 4.80   | 4.50   | 4.50   | 5.50   | 5.10   | 3.30   | 2.90   | 2.90   | 3.20   | Na <sub>2</sub> O |
| K20                            | 0.75   | 0.62   | 2.17   | 1.37   | 0.35   | 1.32   | 1.26   | 0.63   | 0.78   | 0.87   | 0.77   | 1.22   | 1.30   | к <sub>2</sub> 0  |
| P205                           | 0.21   | 0.07   | 0.63   | 0.17   | 0.12   | 0.02   | 0.09   | 0.15   | 0.10   | 0.24   | 0.20   | 0.16   | 0.32   | P205              |
| Loss                           | 0.95   | 2.23   | 0.49   | 0.72   | 1.55   | 1.47   | 1.77   | 2.68   | 7.51   | 0.89   | 0.66   | 1.73   | 0.97   | Loss              |
| Total                          | 99.73  | 99.31  | 99.83  | 99.80  | 99.46  | 99.77  | 101.38 | 99.59  | 100.21 | 99.81  | 101.13 | 99.84  | 99.80  | Total             |
| s                              | 1250   | 3030   | 970    | 280    | 250    | 1020   | 350    | 1640   | 240    | 880    | 810    | 750    | 800    | s                 |
| Ga                             | 19     | 1.7    | 24     | 24     | 0      | 18     | 15     | 16     | 19     | 23     | 21     | 19     | 27     | Ga                |
| Cr                             | 30     | 41     | 22     | 139    | 222    | 32     | 233    | 99     | 152    | 191    | 296    | 361    | 248    | Cr                |
| Ni                             | 7      | 35     | 19     | 104    | 154    | 46     | 93     | 69     | 137    | 94     | 128    | 285    | 152    | Ni                |
| Cu                             | 38     | 253    | 15     | 11     | 0      | 8      | 79     | 63     | 3      | 56     | 137    | 6      | 42     | Cu                |
| Zn                             | 116    | 69     | 107    | 103    | 0      | 107    | 90     | 126    | 56     | 119    | 74     | 229    | 114    | Zn                |
| Rb                             | 8      | 9      | 43     | 22     | 6      | 16     | 18     | 7      | 11     | 10     | 7      | 21     | 16     | RЬ                |
| Sr                             | 250    | 225    | 89     | 221    | 216    | 181    | 353    | 305    | 228    | 315    | 474    | 257    | 561    | Sr                |
| ¥                              | 29     | 14     | 30     | 43     | 16     | 12     | 7      | 21     | 19     | 39     | 20     | 13     | 29     | Y                 |
| Zr                             | 109    | 67     | 393    | 220    | 79     | 43     | 50     | 77     | 86     | 238    | 213    | 88     | 308    | Zr                |
| Nb                             | 2      | 0      | 37     | 6      | 0      | 2      | 4      | 2      | 1      | 6      | 7      | 2      | 7      | Nb                |
| Ва                             | 155    | 137    | 619    | 305    | 146    | 139    | 152    | 185    | 209    | 183    | 134    | 281    | 186    | Ba                |
| La                             | 6      | 2      | 59     | 18     | 2      | 5      | 4      | 5      | 4      | 3      | 6      | 6      | 8      | La                |
| Ce                             | 47     | 40     | 165    | 88     | 42     | 37     | 41     | 46     | 39     | 60     | 71     | 50     | 81     | Ce                |
| Рb                             | 8      | ' 9    | 16     | 8      | 8      | 6      | 9      | 7      | 7      | 8      | 5      | 7      | 8      | Pb                |
| Th                             | 0      | 1      | 22     | 6      | 1      | 0      | 1      | 3      | 3      | 1      | 3      | 0      | 3      | Th                |
| NC                             | 834669 | 789649 | 831678 | 790621 | 797650 | 827689 | 827698 | 774654 | 801654 | 823648 | 831674 | 802642 | 829694 | NC                |

Table VII. The Strathy complex amphibolites and the Armadale ultra-mafic body.

| SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> | 55.77<br>0.11<br>2.46<br>5.72 | 52.63<br>0.61<br>16.09 | 51.55<br>0.55 | BS124<br>50.85 | BS 20<br>50.59 | BS116         | BS219  | BS 9   | BS 55  | BS 32  | BS134  | BS 44  | BS110  | BS 49  |                                       |
|--|-------------------------------|------------------------|---------------|----------------|----------------|---------------|--------|--------|--------|--------|--------|--------|--------|--------|---------------------------------------|
| TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub>                  | 0.11                          | 0.61                   |               | 20.02          |                |               | 50.09  | 49.68  | 49.54  | 48.91  | 48.21  | 48.01  | 46.23  | 45.37  | $sio_2$                               |
| A1203  | 2.46                          |                        |               | 1.17           | 0.27           | 50.50<br>0.67 | 0.59   | 0.52   | 1.48   | 0.66   | 2.01   | 0.58   | 2.52   | 1.64   | TiO2                                  |
|  |                               |                        |               | 14.75          | 15.66          |               | 14.70  | 13.95  | 18,54  | 13.80  | 14.64  | 16.23  | 15.08  | 13.72  | 4.                                    |
|  | 5.72                          |                        | 16.67         |                |                | 15.46         |        |        |        | 10.35  | 13.21  | 11.42  | 12.88  | 10.44  | Al <sub>2</sub> O <sub>3</sub>        |
| Fe <sub>2</sub> O <sub>3</sub>                                   |                               | 10.40                  | 9.85          | 10.78          | 8.08           | 13.21         | 9.53   | 24.04  | 10.13  |        |        | 0.15   | 0.21   | 0.01   | Fe <sub>2</sub> 0 <sub>3</sub><br>Mn0 |
| MnO  | 0.00                          | 0.01                   | 0.17          | 0.00           | 0.17           | 0.16          | 0.00   | 0.13   | 0.04   | 0.35   | 0.16   |        |        |        |                                       |
| MgO  | 20.25                         | 7.45                   | 6.79          | 8.33           | 9.32           | 8.43          | 10.29  | 7.08   | 3.56   | 10.33  | 7.12   | 7.87   | 7.36   | 11.41  | MgO                                   |
| Ca0  | 13.39                         | 4.89                   | 7.69          | 8.80           | 9.00           | 5.40          | 8.26   | 2.59   | 5.45   | 9.16   | 10.72  | 9.04   | 9.68   | 10.99  | CaO                                   |
| Na <sub>2</sub> 0  | 0.90                          | 5.10                   | 4.70          | 3.30           | 4.20           | 4.00          | 3.80   | 1.90   | 6.00   | 3.30   | 2.80   | 3.40   | 3.30   | 1.90   | Na <sub>2</sub> O                     |
| к <sub>2</sub> 0   | 0.20                          | 1.09                   | 0.88          | 0.49           | 0.74           | 1.85          | 0.52   | 0.36   | 1.82   | 0.65   | 0.76   | 1.53   | 1.24   | 1.10   | K20                                   |
| P205   | 0.03                          | 0.14                   | 0.05          | 0.19           | 0.04           | 0.10          | 0.11   | 0.08   | 1,12   | 0.09   | 0.22   | 0.08   | 0.35   | 0.58   | P205                                  |
| Loss   | 1.21                          | 1.52                   | 2.01          | 0.95           | 1.82           | 1.05          | 2.23   | 1.02   | 1,74   | 2.27   | 0.78   | 1.55   | 0.93   | 2.39   | Loss                                  |
| Total  | 100.04                        | 99.93                  | 100.91        | 99.61          | 99.90          | 100.92        | 100.12 | 101.35 | 99.42  | 99.87  | 100.61 | 99.86  | 99.78  | 99.55  | Total                                 |
| s  | 170                           | 160                    | 170           | 1000           | 400            | 740           | 300    | 770    | 2620   | 460    | 340    | 590    | 990    | 1980   | S                                     |
| Ga   | 0                             | 22                     | 16            | 21             | 14             | 18            | 17     | 27     | 27     | 15     | 24     | 17     | 27     | 18     | Ga                                    |
| Cr   | 2168                          | 172                    | 68            | 266            | 540            | 295           | 540    | 654    | 19     | 476    | 205    | 88     | 176    | 578    | Cr                                    |
| Ni   | 1784                          | 106                    | 70            | 57             | 298            | 123           | 336    | 210    | 3      | 356    | 102    | 53     | 100    | 394    | Ni                                    |
| Cu   | 0                             | 0                      | 0             | 31             | 18             | 144           | 2      | 0      | 41     | 0      | 37     | O      | 111    | 16     | Cu                                    |
| Zn   | 0                             | 102                    | 270           | 101            | 94             | 77            | 135    | 61     | 115    | 223    | 100    | 97     | 122    | 108    | Zn                                    |
| Rb   | 1                             | 21                     | 10            | 4              | 8              | 7             | 6      | 7      | 41     | 12     | 16     | 16     | 15     | 12     | Rb                                    |
| Sr   | 187                           | 124                    | 283           | 314            | 297            | 474           | 144    | 17     | 1310   | 243    | 233    | 291    | 369    | 517    | Sr                                    |
| Y  | 2                             | 14                     | 18            | 23             | 7              | 19            | 13     | 13     | 32     | 15     | 34     | 11     | 33     | 28     | Y                                     |
| Zr   | 26                            | 67                     | 83            | 153            | 45             | 214           | 65     | 58     | 860    | 98     | 217    | 35     | 318    | 276    | Zr                                    |
| Nb   | 2                             | 2                      | 4             | 3              | 2              | 6             | 0      | 1      | 31     | 0      | 5      | 7      | 13     | 17     | Nb                                    |
| Ва   | 68                            | 126                    | 285           | 230            | 180            | 131           | 112    | 86     | 541    | 117    | 205    | 157    | 158    | 237    | Ва                                    |
| La   | 4                             | 4                      | 4             | 10             | 4              | 2             | 4      | 1      | 55     | 1      | 3      | 1      | 5      | 45     | La                                    |
| Ce   | 49                            | 46                     | 44            | 66             | 37             | 69            | 49     | 35     | 174    | 55     | 52     | 49     | 75     | 155    | Ce                                    |
| Pb   | 4                             | 5                      | 11            | 8              | 10             | 6             | 8      | 7      | 12     | 14     | 7      | 9      | 8      | 4      | Pb                                    |
| Th   | 1                             | 0                      | 0             | 5              | 0              | 0             | 0      | 0      | 4      | 1      | 2      | 0      | 0      | 7      | Th                                    |
| NC   | 778651                        | 796658                 | 623687        | 833668         | 795649         | 830678        | 802661 | 796636 | 824691 | 805662 | 815688 | 829695 | 831681 | 827697 | NC                                    |

BS289 is the Armadale ultra-mafic body.

Table VIII. Marbles from Fort Mor (BS), Glenelg (GL), Shin (LS), Urquhart (UQ) and along the Sgurr Beag slide near Dornie (W).

|                                | BS 17  | BS 24  | BS 16  | GL 8     | GL 7   | GL 3   | GL 4     | LS 5     | LS 4     | UQ 4B   | UQ 4C  | w 9A   | W 9B   |                                |
|--------------------------------|--------|--------|--------|----------|--------|--------|----------|----------|----------|---------|--------|--------|--------|--------------------------------|
|                                |        |        |        |          |        |        |          |          |          |         |        |        |        |                                |
| SiO <sub>2</sub>               | 59.06  | 50.86  | 37.03  | 13.42    | 12.59  | 12.46  | 12.03    | 40.80    | 23.45    | 51.60   | 40.04  | 14.03  | 11.14  | SiO <sub>2</sub>               |
| TiO <sub>2</sub>               | 0.53   | 0.77   | 0.88   | 0.12     | 0.14   | 0.05   | 0.04     | 0.02     | 0.02     | 0.38    | 0.31   | 0.01   | 0.01   | TiO2                           |
| A1 <sub>2</sub> 0 <sub>3</sub> | 15.06  | 14.02  | 16.91  | 2.18     | 2.13   | 0.76   | 0.75     | 0.45     | 0.39     | 7.88    | 8.71   | 0.16   | 0.12   | A1203                          |
| Fe <sub>2</sub> 0 <sub>3</sub> | 3.73   | 6.29   | 8.16   | 2.57     | 2.58   | 1.36   | 1.20     | 1.08     | 1.16     | 0.97    | 2.01   | 0.46   | 1.18   | Fe <sub>2</sub> 0 <sub>3</sub> |
| MnO                            | 0.00   | 0.00   | 0.09   | 0.06     | 0.07   | 0.05   | 0.04     | 0.00     | 0.01     | 0.00    | 0.00   | 0.00   | 0.00   | MnO                            |
| Mg0                            | 2.84   | 4.68   | 4.37   | 20.08    | 18.31  | 19.72  | 20.89    | 14.08    | 9.07     | 2.97    | 9.76   | 20.83  | 20.98  | MgO                            |
| Ca0                            | 14.32  | 16.61  | 18.89  | 30.81    | 30.96  | 30.13  | 31.81    | 31.84    | 43.62    | 19.18   | 26.16  | 32.19  | 32.48  | Ca0                            |
| Na <sub>2</sub> 0              | 2.90   | 3.00   | 1.00   | 0.00     | 0.00   | 0.00   | 0.00     | 0.00     | 0.00     | 1.70    | 0.70   | 0.00   | 0.00   | Na <sub>2</sub> 0              |
| K <sub>2</sub> O               | 0.41   | 0.72   | 1.40   | 0.09     | 0.03   | 0.25   | 0.17     | 0.03     | 0.03     | 1.65    | 0.16   | 0.01   | 0.01   | к20                            |
| P2O5                           | 0.15   | 0.14   | 0.29   | 0.03     | 0.04   | 0.02   | 0.02     | 0.02     | 0.08     | 0.07    | 0.10   | 0.03   | 0.02   | P2O5                           |
| Loss                           | 1.76   | 2.69   | 10.43  | 30.80    | 33.25  | 34.00  | 32.95    | 11.78    | 22.20    | 11.84   | 11.64  | 32.18  | 33.00  | Loss                           |
| Total                          | 100.77 | 99.78  | 99.45  | 100.31   | 100.10 | 98.80  | 100.90   | 100.10   | 100.03   | 98.24   | 99.59  | 99.90  | 98.94  | Total                          |
| s                              | 1710   | 1670   | 2140   | 210      | 330    | 0      | 310      | 120      | 260      | 600     | 80     | 610    | 5390   | S                              |
| Ga                             | 19     | 23     | 46     | 5        | 5      | 2      | 3        | 4        | 4        | 10      | 12     | 2      | 2      | Ga                             |
| Cr                             | 60     | 93     | 56     | 36       | 37     | 24     | 27       | 26       | 20       | 39      | 37     | 21     | 22     | Cr                             |
| Ni                             | 47     | 66     | 21     | 17       | 20     | 2      | 7        | 7        | 1        | 13      | 12     | 0      | 2      | Ni                             |
| Cu                             | 24     | 36     | 4      | 4        | 41     | 28     | 28       | 0        | 0        | 0       | 0      | 3      | 132    | Cu                             |
| Zn                             | 81     | 109    | 67     | 27       | 27     | 13     | 7        | 55       | 42       | 14      | 32     | 5      | 8      | Zn                             |
| Rb                             | 6      | 11     | 49     | 2        | 1      | 8      | 5        | 1        | 0        | 27      | 3      | 0      | 1      | Rb                             |
| Sr                             | 976    | 1119   | 1696   | 98       | 111    | 97     | 82       | 112      | 198      | 907     | 1248   | 138    | 145    | Sr                             |
| Y                              | 20     | 23     | 34     | 7        | 9      | 1      | 1        | 3        | 2        | 9       | 7      | 0      | 0      | Y                              |
| Zr                             | 168    | 306    | 342    | 22       | 31     | 22     | 15       | 10       | 16       | 580     | 383    | 9      | 15     | Zr                             |
| Nb                             | 13     | 13     | 13     | 3        | 5      | . 0    | 2        | 0        | 2        | 6       | 9      | 3      | 6      | Nb                             |
| Ва                             | 279    | 906    | 278    | 54       | 31     | 116    | 67       | 38       | 3        | 463     | 88     | 38     | 35     | Ва                             |
| La                             | 41     | 34     | 34     | 11       | 11     | 5      | 7        | 4        | 4        | 10      | 10     | 2      | 3      | La                             |
| Ce                             | 135    | 111    | 143    | 94       | 107    | 77     | 73       | 58       | 99       | 54      | 75     | 67     | 72     | Ce                             |
| РЬ                             | 20     | 21     | 34     | 3        | 2      | 3      | 2        | 20       | 37       | 9       | 12     | 8      | 9      | Pb                             |
| Th                             | 24     | 16     | 16     | 2        | 1      | 0      | 0        | 2        | 1        | 5       | 5      | 0      | 1      | Th                             |
| NC                             | 774655 | 774654 | 775656 | NG875254 | 875254 | 875254 | 875254 1 | NC526137 | 526137 N | H491304 | 491304 | 008137 | 008137 | NH                             |

The high K/Rb ratios and low Ba values of the Strathy rocks differentiate them from the Moine gneisses in a remarkable manner. On fig. 4, the Strathy gneisses occupy a low-Ba, high-K/Rb field with a marked negative correlation, whereas the Moine gneisses fall in a high-Ba, low-K/Rb field with a positive correlation (see later).

Many of the geochemical features of the Strathy rocks considered so far have been analogous to those of Lewisian basement orthogneisses from elsewhere in the Sutherland Moine (Moorhouse. 1976; Moorhouse and Moorhouse, 1977, and in prep.) and the foreland Lewisian (Sheraton et al., 1973; Holland and Lambert, 1975). One of the best geochemical discriminants between Moine cover and Lewisian basement rocks is a CaO vs. Y diagram (Moorhouse, 1976; Moorhouse and Moorhouse, 1977, and in prep.). However, the Strathy siliceous gneisses are quite dissimilar from most of the Lewisian gneisses in having low CaO and quite high Y. Therefore, in terms of CaO and Y there is a complete overlap between the Strathy complex gneisses and the Moine gneisses. Although the Strathy gneiss Y levels are approximately

equivalent to the Moine gneisses and higher than the Leverburgh and east Greenland metasediments (Table I), their La and Ce contents are much lower than these rock groups. Ce is equal to the average Scourian from Assynt (Table I) and La is even lower than the average for the Scourian assemblage (26 ppm, Holland and Lambert, 1975). Thus, on a CaO vs. La diagram (fig. 5) the Strathy gneisses fall in a similar field to Lewisian basement orthogneisses from elsewhere (cf. fig. 5 and Moorhouse and Moorhouse, 1977, fig. 4).

On a Y-Ce-La triangular plot (fig. 6) most of the Strathy gneisses fall in a completely separate field from the Moine gneisses. Thus the Strathy gneisses are fundamentally different from the local Moine gneisses and quite dissimilar from normal Archaean gneisses which tend to be Y-poor and La-Ce rich (Tarney, 1976). However, these latter rocks are often considerably more mafic than the Strathy gneisses and as Wright *et al.* (1973) show there is a tendency for the most mafic Archaean granulites to have the highest Ce and La values. Therefore, the differences in Ce and La values could be ascribed to differences in the gross geochemistry

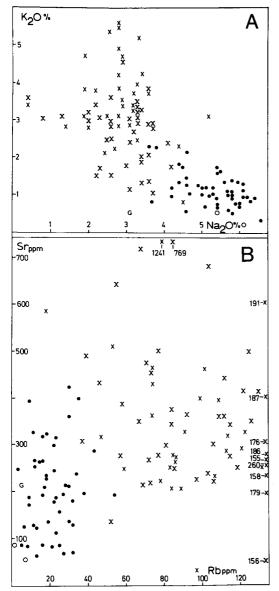


Fig. 3. (A) K<sub>2</sub>O vs. Na<sub>2</sub>O. (B) Sr vs. Rb. Key as fig. 2.

of the rock groups. However, the most silicic Archaean gneisses have much lower Y contents than the Strathy gneisses (Tarney and Windley, 1977) as do many Archaean acid igneous rocks (Arth and Hanson, 1975). In terms of Y level the Strathy rocks appear to be more comparable with younger central European granulites (Tarney and Windley, 1977) than with Archaean rocks.

The Strathy complex hornblende gneisses and amphibolites. Whilst discriminatory diagrams using

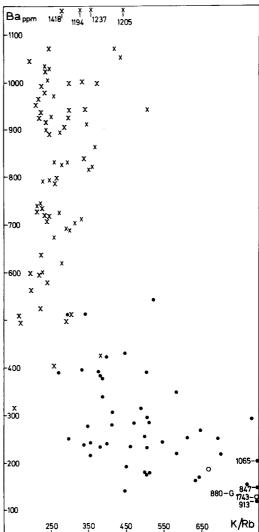


Fig. 4. Ba vs. K/Rb ratio. Key as fig. 2.

SiO<sub>2</sub>, Fe<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> cannot completely separate these basic rocks from the Bettyhill and Kirtomy Moine amphibolites there are significant differences. A comparison of the analyses of the Strathy hornblendic rocks (Tables VI and VII) with Moine amphibolites from the Bettyhill and Kirtomy assemblages (Moorhouse and Moorhouse, 1979a, Table VI, analyses BS277-BS195 Bettyhill, BS69-BS100 Kirtomy amphibolites), and a comparison of the average analyses (Table III) indicate that the Strathy rocks are poorer in TiO<sub>2</sub>, t·Fe<sub>2</sub>O<sub>3</sub>, K<sub>2</sub>O, Rb, Y, Zr, Ba, and richer in Na<sub>2</sub>O.

Even allowing for the fact that Moorhouse and Moorhouse (1979a) demonstrated that the amphi-

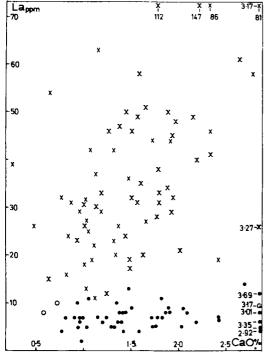


Fig. 5. La vs. CaO. Key as fig. 2.

bolites in the Sutherland Moine are geochemically quite varied, the Strathy basic rocks as a whole fall outside this variation and appear to have a different origin. For example the Moine amphibolites fall into two groups on an AFM plot (Moorhouse and Moorhouse 1979a, fig. 14) one with a tholeiitic trend, the other with a mild alkaline trend. However the Strathy basic rocks show a quite different calc-alkaline trend (fig. 7).

The Port Mor marble and calc-silicate rocks. Three examples of Port Mor diopside-scapolite marble with minor amounts of calcite have been analysed (Table VIII). Two quartz-plagioclase-clinozoisite mica, and two quartz-plagioclase-diopside-garnet calc-silicates, from the Betty-hill and Kirtomy Moine respectively, have been analysed for comparison (Table V), as have two calc-silicate rich clinopyroxene marbles from Glen Urquhart, four calcite-rich forsteritic marbles from the Glenelg Lewisian, two calcite-rich forsteritic-marbles from a Lewisian slice along the Sgurr Beag slide, and two calcite-rich clinopyroxene marbles from the Lewisian slice at the Aird of Shin (Table VIII).

It is evident that the gross geochemistry of these rocks (see Tables V and VIII) is controlled by their calcite/silicate ratio. Obviously the geochemical

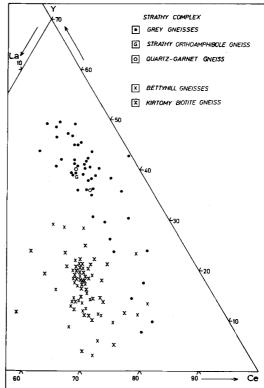


Fig. 6. Y-La-Ce plot.

discrimination of marbles/calc-silicates with different origins using diagrams such as CaO vs. MgO (Winchester and Lambert, 1970) is unlikely to be satisfactory, especially as these rocks are prone to metasomatic interactions. Here the mineralogy of

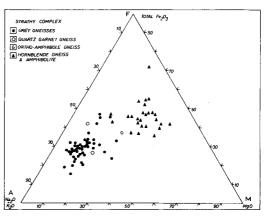


Fig. 7. A (Na<sub>2</sub>O + K<sub>2</sub>O)-F (total Fe as Fe<sub>2</sub>O<sub>3</sub>) M (MgO) plot for Strathy gneisses and amphibolites.

the Port Mor marbles is quite different from the silicate-rich Moine calc-silicates, which are therefore geochemically distinct from the marbles. The Port Mor diopside-scapolite marble geochemistry is most closely paralleled by the calcite-poor, pyroxene-bearing Glen Urquhart rocks.

Discussion of the evolution of the Strathy complex

Mineralogical considerations. Sheraton et al. (1973) concluded that the Scourian rocks of Assynt acquired their low K<sub>2</sub>O and Rb geochemical character by a process of 'charnockitization' of substantially diopside-normative rocks, producing granulite-facies calcic-plagioclase hypersthenic gneisses. This process cannot be responsible for the geochemical peculiarities of the Strathy complex (Tables I and II) as here 25% of the gneisses are corundum-normative and 50% of the gneisses which show diopside in the norm have only extremely small proportions, commonly less than 1%. In addition, the plagioclase in the Strathy rocks is Na-rich, quite unlike that of the Assynt rocks, and there is no unequivocal evidence of granulite-facies mineral parageneses.

The earliest mineral parageneses still recognizable in the Strathy complex contain, in addition to dominant quartz and oligoclase, low-Ca amphibole (or hornblende in the most calcic gneisses), garnet, and occasional staurolite; this assemblage would be stable in the upper amphibolite facies (Turner, 1968) and may have been produced by reactions such as:

1. chlorite+muscovite → staurolite+biotite+garnet+H<sub>2</sub>O.

Ca-poor amphibole by a reaction such as

2. biotite + quartz  $\rightarrow$  K-feldspar + orthoamphibole + H<sub>2</sub>O.

Hornblende in the hornblende-bearing gneisses: 3. biotite + plagioclase + quartz  $\rightarrow$  K-feldspar + albite + hornblende.

These last two reactions are known to take place under the highest amphibolite-facies conditions (Knabe, 1966), but the fact that, in the Strathy complex, staurolite was apparently stable in the presence of quartz may indicate very high pressure, i.e. lower crustal conditions (Schreyer and Seifert, 1969). Thus it is quite possible that any original K-feldspar component produced in reactions such as 2 and 3 was removed during a period of early anatexis. Following this was lower grade, but probably still amphibolite-facies, retrogression producing biotite-bearing assemblages, and 'blebby' quartz (by reversing reactions 2 and 3) which overprints the quartz-plagioclase polygonal groundmass. Then prograde metamorphism to upper amphibolite-facies sillimanite grade, involving the production of garnet, spinel, and sillimanite,

which also affected the surrounding Moine, produced trondhjemites in the Strathy complex and granites in the Moine rocks by anatexis of the respective gneisses (Moorhouse, 1979; Moorhouse and Moorhouse, in prep.).

Geochemical considerations. With regard to the above, the evidence of the Ba vs. K/Rb ratio diagram (fig. 4) is of great importance. The positive correlation shown by the Moine gneisses on this plot can be explained by the removal of anatectic granite from the biotite-rich (Rb-accepting) Moine gneisses. On average, the relatively biotite-free granite differs from the parent Kirtomy gneisses (Table I, column E), in containing less Rb (average 81 ppm) as this is held in the biotite of the gneiss, much more Ba (average 2480 ppm) and more K<sub>2</sub>O (average 4.49 %, Moorhouse, 1979; Moorhouse and Moorhouse, in prep.) as these go into the Kfeldspar of the melt (Ba being preferentially accepted relative to Rb). Thus those Moine gneisses which have supplied least anatectic melt will tend to have the highest Ba and K/Rb ratios, whereas those gneisses which have supplied the most anatectic granite have the lowest Ba and K/Rb ratios, having lost proprtionately more of their K<sub>2</sub>O and Ba than their Rb. Thus, the Moine gneisses have a positive Ba-K/Rb correlation.

This is quite distinct from the negative correlation shown by the Strathy gneisses, which is simply explained by postulating that anatectic melts have also been extracted from them, probably during two distinct phases of metamorphism but under rather different conditions. During the earliest phase any original K-feldspar in the rock and the K-feldspar component produced by reactions 2 and 3 went into an anatectic melt which was removed from the complex. Rb and to a lesser extent Ba, would also be concentrated in the melt as there is no Rb accepting phase stable in the gneisses once biotite has been replaced by amphibole (reactions 2 and 3), whereas plagioclase may contain significant amounts of Ba (Taylor, 1965). The average gneiss composition after removal of this first anatectic melt but prior to the later anatectic phase is crudely approximated by a weighted average of 60% siliceous gneiss plus 40% trondhjemite, these being the approximate proportions seen in the field (Table II, column F). This indicates that prior to the later anatectic phase the Strathy gneisses were depleted in K<sub>2</sub>O, Rb, and Ba relative to highgrade terrains such as the Brazilian and Musgrave granulites (Table II) but had similar Sr contents; which is reflected in the higher K/Rb and lower Rb/Sr ratios of this Strathy, 'post-early anatexis' gneiss, mean analysis (Table II, column F).

Therefore, during the early anatexis of the Strathy gneisses we believe significant proportions

of the original  $K_2O$ , Rb, and Ba were completely removed from the complex via an anatectic melt, whilst Sr remained relatively immobile in the Ca-bearing plagioclase and amphibole. During the later anatectic phase Sr and Ba are strongly concentrated in the plagioclase-rich trondhjemite (Table II, column G),  $K_2O$  and Rb are only slightly enriched as by this stage retrogression of amphibole had produced biotite in the Strathy gneisses which would tend to hold the  $K_2O$  and Rb.

The Y-Ce-La distribution. A similar line of reasoning explains the relative abundances of Y, La, and Ce in the Strathy rocks. Of the RE-holding minerals those with small lattice sites, such as garnet and apatite, show preferential enrichment in the heavy RE (Gd-Lu) and Y (Taylor, 1965; Frey, 1969), whereas the light RE tend to follow the large elements Sr, Ba, and K.

Both garnet and apatite are common in the Strathy rocks, and the garnet certainly was stable in the gneisses during both periods of anatexis. Therefore Y would tend to be retained in the Strathy rocks whereas Ce and La would have a tendency to be lost from the complex in the proposed early anatectic phase. Thus this would explain why the Ce and La content of the Strathy complex is low, lower than the Moine gneisses and the Leverburgh and East Greenland metasediments (Table I). It is also worth noting that Moorhouse and Moorhouse (1977) found that Y was immobile, but La and Ce were mobile and concentrated in the felsic fraction, during migmatization of certain Lewisian rocks.

# Evolution of the Strathy complex in comparison with other terrains

In contrast to this proposal for the geochemical evolution of the Strathy complex, various authors have considered that anatexis of metavolcanics and metasediments was not primarily responsible for the removal of  $K_2O$  etc. from high-grade terrains such as the Assynt gneisses. For example, Sheraton et al. (1973) suggest the removal of  $K_2O$  and Rb via an aqueous fluid whilst others, notably Holland and Lambert (1975) and Tarney and Windley (1977) appeal to mechanisms involving the direct derivation of material from the mantle.

The high Ba and La content of the Assynt granulites/gneisses (Sheraton et al., 1973) does militate against the formation of anatectic melts as this would be likely to remove substantial Ba and La. Sheraton et al. also consider that 'K-feldspar should be a stable phase in rocks which are not diopside normative', this is manifestly not the case in the Strathy complex for the reasons given above.

Cooper and Field (1977), when considering the origin of the Tromøy granulites state that in spite of these rocks having Ba and Sr values lower than many other granulites the Tromøy rocks are 'much too acid for there to have been major amounts of granite removed'. However, it is an observed fact that when an anatectic melt is removed from acid quartz-rich rocks the residue becomes richer in silica, i.e. more acid (Winkler, 1967; and a consideration of the Qz-Ab-An-Or-H<sub>2</sub>O system).

The parental material of the Strathy complex. These gneisses were obviously originally rich in silica, whatever processes are considered to have subsequently modified them. Nevertheless, the original parental material was probably not as rich in  $SiO_2$  as the present day siliceous gneiss (Table I, column A) or the gneiss-trondhjemite mean (Table II, column F), whilst it would have been richer in  $K_2O$ , Rb, plus Ba, La, and Ce to a lesser extent.

In view of the fact that probable calcareous metasediments occur in the Strathy complex (Port Mor marble and Strathy calc-silicates) plus possible quartz-rich metasediments (quartz-magnetite gneiss, quartz-garnet gneiss) could not a large proportion of the Strathy complex rocks represent paragneiss with inherited geochemical peculiarities? This would appear unlikely because, for example, on the SiO<sub>2</sub>-TiO<sub>2</sub> diagram (fig. 2A) the Strathy gneisses occupy a position distinct from typical metasediments (e.g. the Moine gneisses) but fall in a field occupied by calc-alkali igneous rocks and Archaean gneisses (Tarney, 1976). Secondly, when the Strathy gneisses and amphibolites are plotted on an AFM diagram (fig. 7) they display a calcalkali trend very similar to the trend of Archaean gneisses from Scotland and Greenland (Sheraton et al., 1973). Finally, when plotted on variation diagrams such as MgO vs. Ni and Cr (figs. 8A, B) a trend is apparent running from the siliceous gneisses through the hornblende gneisses and the amphibolites, like trends observed in many igneous sequences.

It is possible that whilst the basic rocks are of igneous origin, the more acid are paragneisses or volcano-clastics derived from a calc-alkaline sequence. However, as the mafic minerals containing the bulk of the Cr and Ni are the first to break down in the sedimentary process we would expect the above trends to be less apparent. Therefore we believe that the bulk of the Strathy complex is probably not of metasedimentary derivation and the most likely parent for the siliceous grey gneisses would seem to be an acid volcanic or its plutonic equivalent. In view of the foregoing discussion, use of Winchester and Floyd's (1977) magma-type discrimination which indicates a rhyolitic origin is

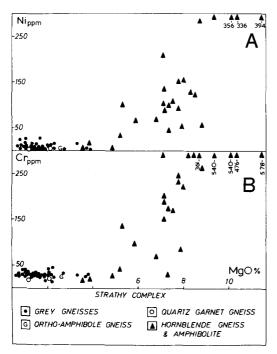


Fig. 8. (A) Na vs. MgO. (B) Cr vs. MgO. Strathy gneisses and amphibolites.

certainly misleading. The actual parent material was probably less SiO<sub>2</sub>-rich but is not likely to have been as basic as andesite, if only because this would probably have resulted in a diopside-normative 'charnockitization' process leading to hypersthene-bearing granulite (Sheraton *et al.*, 1973). Therefore, a low-CaO dacitic, perhaps inclining towards Na-rich or pantelleritic, composition appears to be the most likely parental material.

Minor amounts of metasediments could have become involved with contemporaneous volcanism if the complex was originally formed in an islandarc type environment. If the gneiss precursors were plutonic they may represent a basement upon which the metasediments were deposited unconformably, or an underplating mechanism may have been responsible for the involvement of the sediments in the complex (Tarney and Windley, 1977).

#### Conclusions

The only geochronological work carried out on the Strathy complex to date (Rb-Sr whole rock; Taylor, Moorhouse, and Moorhouse, unpub.) unfortunately proved inconclusive, yielding unreliable indications of a middle Proterozoic event. Thus, in the absence of any reliable geochronological data, although there are numerous possibilities, we believe the available evidence indicates that the Strathy complex is a block of the sub-Moine basement most probably referable to the Lewisian complex. Although an early Scourian origin cannot be ruled out, the REE and Y data suggest a post-Archaean origin. Therefore the complex possibly originated as a Laxfordian, or even early Grenvillian, volcanic supracrustal sequence with minor sediments and intrusives. One of the most notable features of the complex is that although it has not apparently undergone granulite-facies metamorphism it has a geochemistry deficient in 'incompatible' elements. This geochemical imprint could have been acquired by the removal of an anatectic melt during upper amphibolite-facies metamorphism whilst the complex formed part of the lower crust, most probably during the Laxfordian or Grenvillian cycle.

Retrogression during later events has only slightly modified this unusual geochemistry. During the metamorphism of the adjacent Moine cover sequence, H<sub>2</sub>O and probably some K<sub>2</sub>O and Rb were introduced into the complex, producing trondhjemitic anatectic melts simultaneously with the production of granitic anatectic melts in the local Moine gneisses (Moorhouse, 1979). Therefore, if Strathy complex-type basement is present elsewhere beneath the east Sutherland Moine, it may have contributed significant amounts of sodic melt to the east Sutherland migmatite-granite complexes (Brown, 1967, 1971; Moorhouse, 1979, Moorhouse and Moorhouse, in prep.).

The Strathy complex occurs as a fault-bounded block of basement aspect which is clearly defined by a strong positive aeromagnetic anomaly. It is structurally, lithologically, and geochemically distinct from the tectonic inliers of Lewisian basement found elsewhere in the Moine succession and its geochemistry is incompatible with a derivation from the local Moine gneisses. Further research may enable us to recognize other examples of Strathy complex type basement within the Moine but as far as we know at the present time, the Strathy complex is unique within the Northern Highlands of Scotland.

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