

## Niobian titanite and eudialyte from the Ilomba nepheline syenite complex, north Malawi

DURING a chemical and mineralogical study of the Ilomba nepheline syenite complex, northern Malawi, titanites have been found which contain up to 11% by weight of  $\text{Nb}_2\text{O}_5$ . Bloomfield *et al.* (1981) described titanite separated from aegirine pegmatite at Ilomba with 3.22%  $\text{Nb}_2\text{O}_5$ , which they considered to be the highest values of niobium recorded for this mineral. As discussed by Deer *et al.* (1982), the position of Nb in titanite is uncertain but the chemical data presented here indicate that it must replace titanium in the octahedral sites, and thus plays a similar role to the tantalum in a tantalum-rich variety of titanite described by Clark (1974).

The specimen containing the titanite was

collected by Bloomfield from southwest of the summit of Ilomba and is now in the collection of the Natural History Museum, London—number BM 1987, P6(75). The rock is a foliated nepheline syenite comprising perthite, nepheline and aegirine with accessory titanite, eudialyte, zircon, pyrochlore and carbonate. The titanites occur as abundant but small (generally  $<0.1$  mm), discrete crystals.

Analyses were made using a Cambridge Instruments Microscan 9 wavelength-dispersive microanalyser operated at 20 kV with a specimen current of 25 nA. A selection of analyses is given in Table 1.  $\text{Nb}_2\text{O}_5$  values vary from 1.9% to 11.12% by weight, with all the lower values found

Table 1  
Analyses of niobian titanite

$\text{SiO}_2$	29.04	29.12	29.03	28.77
$\text{TiO}_2$	29.51	27.15	26.28	32.17
$\text{ZrO}_2$	1.42	1.58	1.65	1.58
$\text{Nb}_2\text{O}_5$	7.84	9.91	11.12	5.88
$\text{Ta}_2\text{O}_5$	0.15	0.17	0	0.24
$\text{Al}_2\text{O}_3$	0.37	0.39	0.35	0.22
$\text{Fe}_2\text{O}_3$	2.86	3.16	3.12	2.60
MnO	0.11	0.08	0.14	0.14
CaO	26.15	25.33	24.72	26.79
SrO	0.13	0.20	0.16	0.14
$\text{Na}_2\text{O}$	0.71	1.07	1.29	0.89
F	0.40	0.24	0.45	0.25
	98.69	98.40	98.33	99.67
O=F	0.17	0.10	0.19	0.10
Total	98.52	98.30	98.14	99.57
Numbers of ions on the basis of 5 oxygens				
Si	0.997	1.000	1.007	1.007
Al	0.003		-	-
Al	0.012	0.016	0.014	-
$\text{Fe}^{3+}$	0.074	0.082	0.082	0.066
Ti	0.762	0.706	0.687	0.821
Nb	0.122	0.155	0.175	0.090
Ta	0.001	0.002	0.000	0.002
Zr	0.024	0.027	0.028	0.026
Mn	0.003	0.002	0.004	0.004
Na	0.047	0.072	0.087	0.059
Ca	0.962	0.939	0.921	0.974
Sr	0.003	0.004	0.003	0.003
F	0.043	0.026	0.049	0.027

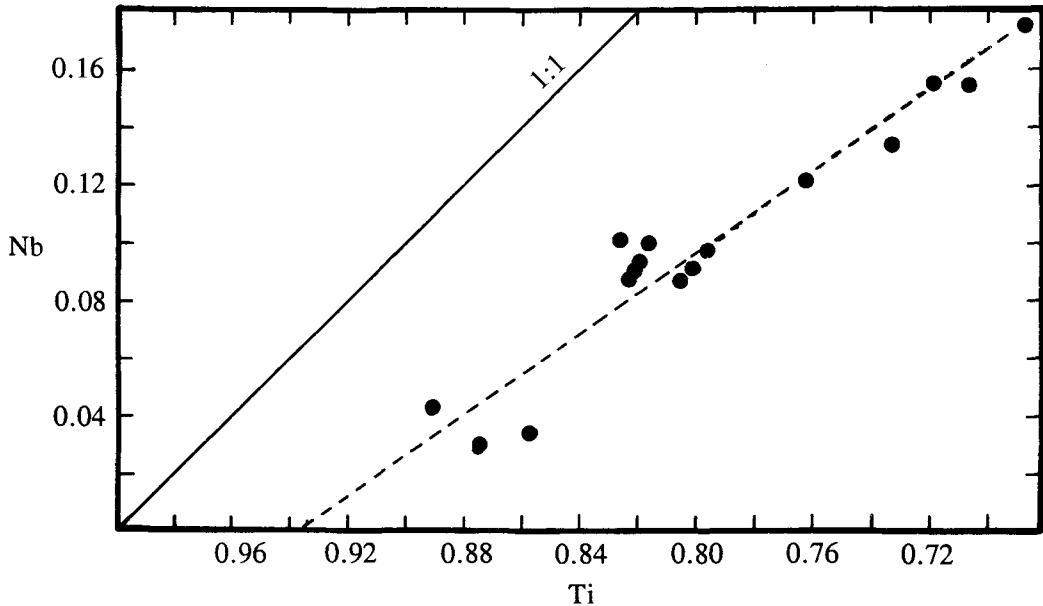


Fig. 1. Plot of Nb against Ti (atoms) for Nb-rich titanite. A strong correlation is apparent. The Nb must occupy the octahedral sites, although there is not sufficient Nb to account for all the deficiency in Ti.

in the cores of crystals and the higher ones close to crystal margins. Tenors of zirconium are also consistently high with values up to 2.67%, while iron, calculated as  $\text{Fe}_2\text{O}_3$ , is always >1% and may be >2.5%. There is a strong correlation between niobium and titanium, illustrated by Fig. 1, indicating that the niobium must be occupying the octahedral sites. Although the titanium clearly decreases as niobium increases there is insufficient niobium alone to fill the octahedral sites, and even after the addition of aluminium and iron this site is still apparently not fully occupied. Although the position of Zr is considered to be uncertain (Deer *et al.*, 1982) its assignment to the Y position, as followed by Sahama (1946), results in a better balance. Following Clark (1974) Ta is also assigned to the Y position.

The presence of such an abundance of Nb in the Ilomba titanite must reflect a high Nb activity in the primary melt, and this is also indicated by the presence of pyrochlore and eudialyte which is unusual in containing >3.5%  $\text{Nb}_2\text{O}_5$ . An analysis of the eudialyte is given in Table 2.

This unusual Nb-rich magmatic environment is reflected in the whole-rock chemistry of the sodalite and nepheline syenites in which Nb/Ta ratios vary from 40 to 60, compared with Nb/Ta ratios of 15–20 typical of most igneous rocks (Woolley *et al.*, in prep.). Such ratios are more

Table 2  
Analysis of niobium-bearing eudialyte

$\text{SiO}_2$	48.34	Number of ions on the basis of 19 oxygens	
$\text{TiO}_2$	0.02		
$\text{ZrO}_2$	11.42		
$\text{Nb}_2\text{O}_5$	3.56	Si	6.288
FeO*	1.75	Ti	0.002
MnO	6.50	Zr	0.724
CaO	15.04		
$\text{Na}_2\text{O}$	8.80	Nb	0.209
$\text{Y}_2\text{O}_3$	0.54	$\text{Fe}^{2+}$	0.190
$\text{La}_2\text{O}_3$	0.74	Mn	0.716
$\text{Ce}_2\text{O}_3$	0.64		
$\text{Pr}_2\text{O}_3$	<0.15	Na	2.219
$\text{Nd}_2\text{O}_3$	<0.12	Y	0.037
$\text{Sm}_2\text{O}_3$	0.07	La	0.036
$\text{Gd}_2\text{O}_3$	<0.15	Ce	0.030
$\text{ThO}_2$	<0.1	Sm	0.000
$\text{UO}_2$	<0.1	Ca	2.096
Cl	2.02		
	99.44	Cl	0.363
O=Cl	0.44		
Total	99.00		

\* all iron determined as FeO.

typical of carbonatites and given the nearby presence of the Nachendzwaya carbonatite complex in Tanzania may suggest a linkage between these intrusions.

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