

Note on Colorado Hydrophane.

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[Read January 22nd, 1889.]

SOME experiments on the hydrophane recently found in Colorado led Mr. G. F. Kunz (*Am. J. Sc.* XXXIV. pp. 479-480, Dec. 1887) to the extraordinary conclusion that this mineral absorbed more than its own bulk of water. Mr. Kunz, with great kindness, furnished me with a supply of this variety of opal from the locality in question. I can confirm the accuracy of his experimental results, but not of his deductions therefrom.

A flat piece of this hydrophane weighing $\cdot 578$ gram absorbed $\cdot 276$ gram of water, becoming nearly transparent. This amount corresponds to an addition of $47\cdot 75$ parts of water to 100 of opal, so that the saturated hydrophane contained $32\cdot 43$ per cent. of water. I determined the volume of the mineral, both in the dry and wet state, by weighing the quantity of pure mercury which it displaced from a small glass cup (with a covering plate) full of that metal. In this way I proved that the volume of the wet hydrophane was identical with that of the dry. The mean weight of mercury displaced in several experiments was $7\cdot 415$ grams. This, corrected for temperature and reduced to its equivalent volume and weight of water, corresponds to $\cdot 546$ cub. cent. or grams, and gives $1\cdot 06$ as the specific gravity of the dry hydrophane.

From the above data—namely, $\cdot 546$ cub. cent. for the volume of the hydrophane, and $\cdot 276$ cub. cent. for the volume of the water it absorbs—we reach the conclusion that 1,000 volumes of hydrophane absorb 505 volumes of water, that is, just over half the original volume of the mineral.

As there is room for the imbibition of $\cdot 276$ cub. cent. of water in the piece of hydrophane with which my experiments were made, we may conclude that the actual opal-substance therein occupies the space of $\cdot 270$ cub. cent. only, out of the total volume of $\cdot 546$. For $\cdot 546 - \cdot 276 = \cdot 270$. We deduce thence the figure $2\cdot 14$ for the specific gravity of the opal-substance free from interstitial air. Mr. Kunz, by the ordinary method of weighing in air and then in water, obtained the figure $2\cdot 122$ for the specimens with which he worked.