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**Crystal Data:** Hexagonal. Point Group:  $\overline{3} 2/m$  or 3m. Crystals typically pseudocubic  $\{01\overline{1}2\}$  or tabular  $\{0001\}$  with flat vicinal rhombohedra, to 1 cm; fibrous to columnar, porcelaneous, commonly granular to dense massive.

**Physical Properties:** Cleavage: On  $\{0001\}$ , perfect. Fracture: Conchoidal. Tenacity: Brittle. Hardness = 3.5-4 D(meas.) = 2.6-2.9 D(calc.) = 2.82 Strongly pyroelectric.

**Optical Properties:** Transparent to translucent. *Color:* Colorless if pure; may be white, pale shades of gray, yellow, red, to reddish brown. *Streak:* White. *Luster:* Vitreous, somewhat pearly on {0001}.

Optical Class: Uniaxial (+).  $\omega = 1.572$   $\epsilon = 1.592$ 

**Cell Data:** Space Group:  $R\overline{3}m$  or R3m (synthetic). a = 6.981(1) c = 17.331(4) Z = 3

**X-ray Powder Pattern:** Marysvale, Utah, USA. 2.89 (100), 2.99 (94), 2.293 (76), 1.926 (66), 4.96 (50), 1.503 (34), 5.77 (26)

Chemistry:

	(1)	(2)
$SO_3$	38.34	38.66
$P_2O_5$	0.58	
$SiO_2$	0.22	
$Al_2O_3$	37.18	36.92
$Na_2O$	0.33	
$\overline{K_2O}$	10.46	11.37
$H_2O$	12.99	13.05
Total	100.10	100.00
UA1(00)(011)		

(1) Marysvale, Utah, USA. (2)  $\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$ .

Polymorphism & Series: Forms a series with natroalunite.

Mineral Group: Alunite group.

**Occurrence:** Formed between 15 °C and 400 °C by the action of sulfate, which may be generated from pyrite or solfataric action, on aluminous rocks, commonly accompanied by kaolinitization and silicification.

Association: Kaolinite, halloysite, diaspore, pyrite, gypsum, quartz.

**Distribution:** Many localities, some with deposits of immense size. From Tolfa, 14 km east-northeast of Civitavecchia, Lazio, Italy. At Mukacheve (Muzijeva) and from Beregavo, near Mukachevo, Ukraine. At Rodalquilar, Almeria Province, Spain. From Décazeville, Aveyron, France. In the USA, in a very large deposit near Marysvale, Ohio district, Piute Co., Utah; in Colorado, at South River, Mineral Co., Red Mountain, Hinsdale Co., in the Rosita Hills, Custer Co., and approximately 100 million t in the Calico Peak porphyry, Rico district, Dolores Co.; at Meiklejohn Mountain, near Beatty, Nye Co., Nevada; in Lassen Volcanic National Park, Shasta Co., California. From Hickory's Pond, near Placentia Bay, Newfoundland, Canada. At Hilton, Bernborough, and other localities in the Mount Isa district, Queensland; from Bullah Delah, New South Wales, Australia.

Name: A contraction of the earlier name aluminilite.

**References:** (1) Palache, C., H. Berman, and C. Frondel (1951) Dana's system of mineralogy, (7th edition), v. II, 556–560. (2) Parker, R.L. (1962) Isomorphous substitution in natural and synthetic alunite. Amer. Mineral., 47, 127–136. (3) Stoffregen, R.E. and C.N. Alpers (1992) Observations on the unit-cell dimensions, H<sub>2</sub>O contents, and  $\delta D$  values of natural and synthetic alunite. Amer. Mineral., 77, 1092–1098. (4) Menchetti, S. and C. Sabelli (1976) Crystal chemistry of the alunite series: crystal structure refinement of alunite and synthetic jarosite. Neues Jahrb. Mineral., Monatsh., 406–417.

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