

The crystal structure of vrbait $\text{Hg}_3\text{Tl}_4\text{As}_8\text{Sb}_2\text{S}_{20}$ *

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Auszug

Die Kristallstruktur von Vrbait wurde bestimmt. Vier Formeleinheiten $\text{Hg}_3\text{Tl}_4\text{As}_8\text{Sb}_2\text{S}_{20}$ sind in der rhombischen Zelle, $a = 13,399$, $b = 23,389$, $c = 11,287$ Å, Raumgruppe $C2ca-C_{2v}^{17}$, enthalten. Die Struktur wurde mittels einer Art Superpositionsmethode auf Grund einer dreidimensionalen Pattersonfunktion gefunden; $R = 4,8\%$.

Die Vrbaitstruktur ist durch unendliche As_2SbS_5 -Ketten parallel c und durch $\text{Hg}_3\text{As}_4\text{S}_{10}$ -Schichten normal zu b charakterisiert. Die ersteren umgeben („sandwich“) die Tl-Atome und bilden Schichtbereiche („slabs“) normal zur b -Achse. Diese Schichtbereiche und die $\text{Hg}_3\text{As}_4\text{S}_{10}$ -Schichten, welche die anderen Tl-Atome umgeben, wechseln in der b -Richtung ab. Die gefundene Struktur erklärt gut die gute Spaltbarkeit parallel (010).

Vrbait ist unseres Wissens die erste Struktur mit gemischten (As,Sb)-Ketten.

Abstract

The crystal structure of vrbait has been determined. Four chemical units of $\text{Hg}_3\text{Tl}_4\text{As}_8\text{Sb}_2\text{S}_{20}$ are contained in the orthorhombic unit cell: $a = 13.399$, $b = 23.389$, $c = 11.287$ Å, symmetry $C2ca-C_{2v}^{17}$. The structure was solved through a kind of superposition method using the three-dimensional Patterson function. The final discrepancy index R is 4.8%.

The structure of vrbait is characterized by infinite As_2SbS_5 chains parallel to c and by $\text{Hg}_3\text{As}_4\text{S}_{10}$ sheets perpendicular to b . The former sandwich the Tl atoms and make slabs perpendicular to the b axis. These slabs and the $\text{Hg}_3\text{As}_4\text{S}_{10}$ sheets, sandwiching the other Tl atoms, are arranged alternately along the b direction. The good cleavage parallel to (010) can be well explained by the structure. Vrbait is the first structure with mixed (As,Sb) chains.

Introduction

Vrbait is a very rare sulfosalt, found by JEŽEK (1912) in a specimen from Allchar, Macedonia. JEŽEK made morphological studies and

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described its point symmetry as $2/m\ 2/m\ 2/m$. KŘEHLÍK (1912) made a chemical analysis of this material and showed its formula to be $\text{TlAs}_2\text{SbS}_5$. The cell dimensions of vrbaite were determined by FRONDEL (1941). He proposed $Cmca$ as the probable space group, taking account of x-ray and morphological data. Recently CAYE, PICOT, PIERROT and PERMINGEAT (1967) reexamined the chemical composition of vrbaite with the aid of electron-probe microanalysis, and obtained a new formula, $\text{Hg}_3\text{Tl}_4\text{As}_8\text{Sb}_2\text{S}_{20}$. This formula was confirmed by NOWACKI (1968).

The present investigation was undertaken to elucidate the coordinations of the sulfur atoms around the thallium atoms which show a quite irregular feature in the structures of sulfosalts, and to examine whether the antimony atoms and the arsenic atoms are in an ordered state or not.

Experimental

A specimen from Allchar, Macedonia (Naturhistorisches Museum, Wien), half of which was being used for chemical analysis (NOWACKI, 1968), was used for the present studies. A spherical crystal with radius 0.106 mm was prepared by BOND's (1951) method for the determination of the cell dimensions and intensities. The lattice constants were obtained from back-reflection Weissenberg photographs calibrated by powder patterns of silicon. A least-squares best fit of the lattice parameters was calculated with the aid of an IBM-1620 program written by N. D. JONES (unpublished). The results agree well with the values obtained by FRONDEL (1941).

Present determination	FRONDEL ¹
$a = 13.399 \pm 0.001 \text{ \AA}$	$13.38 \pm 0.05 \text{ \AA}$
$b = 23.389 \pm 0.001$	23.37 ± 0.05
$c = 11.287 \pm 0.001$	11.25 ± 0.05

The number of formula units, four, was calculated from the cell dimensions obtained and the measured specific gravity, 5.30 (PALACHE, BERMAN and FRONDEL, 1944).

The diffraction symbol for vrbaite is $mmmC_*ca$, which permits $Cmca$ and $C2ca$ as possible space groups. Since a piezoelectric test showed an acentric feature, $C2ca-C_{2v}^{17}$ was selected as the correct space group.

¹ FRONDEL's values are converted to Ångstrom from kX units.

Three-dimensional intensity data (1772 independent reflections) were collected by an automated Weissenberg counter-diffractometer (Supper-Pace) using $\text{CuK}\alpha$ radiation. One hundred and twenty-four of these reflections [with $I < 2.33 \sigma(I)$] were assigned to unobserved reflections.

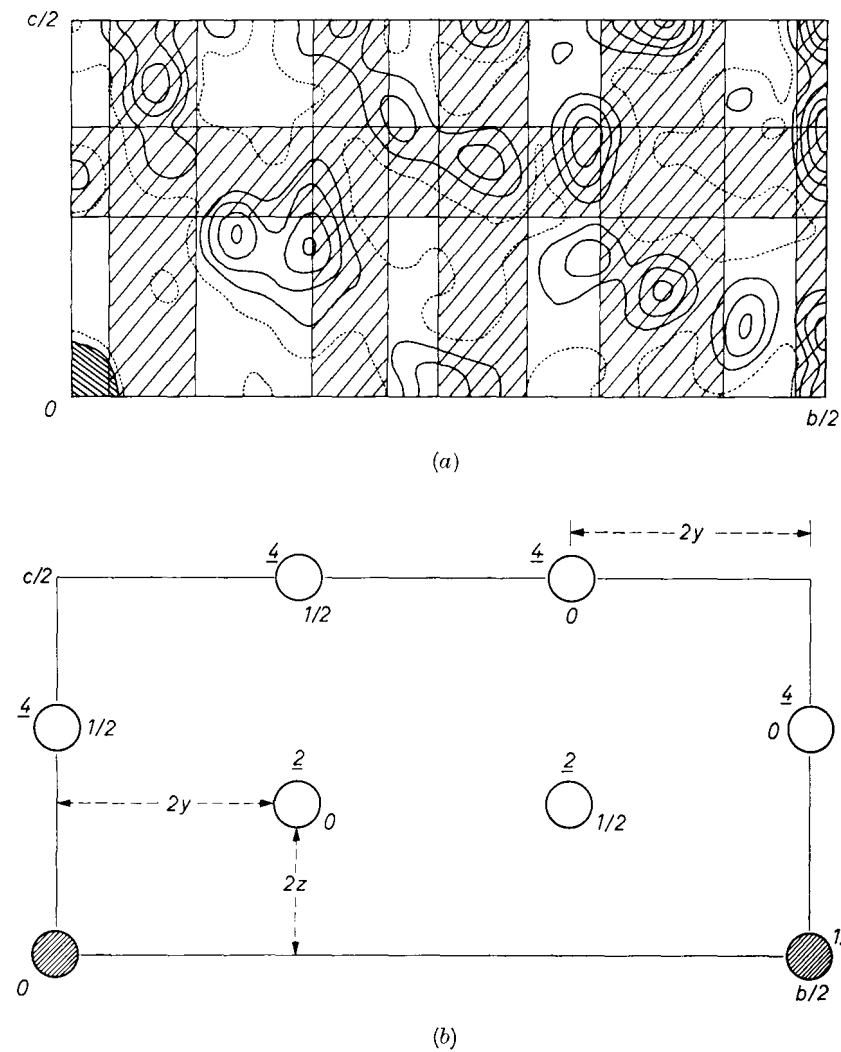


Fig. 1. (a) Patterson section $P(0, v, w)$. Rotation peaks should not exist in the shaded area. Contours are equal but in arbitrary intervals. (b) The vector set for equipoints $8b$. The underlined values indicate weights of points

Determination of the structure and refinement

The Patterson function $P(u,v,w)$, evaluated with the three-dimensional data, showed that all peaks are distributed on the sections $u = n/8$ ($n = 0, 1$ and 2). All heavy peaks which correspond to the vectors between heavy atoms² are on the sections $u = 0$ and $1/4$. These features of the diagram suggest that all atoms are arranged nearly on planes at intervals of $1/8$ along the a direction, and all heavy atoms on every other plane. The only equipoints of the space group $C2ca$ are $4a$ (on twofold axes) and $8b$ (points in the general position)³.

Since there are twelve mercury atoms in the cell and since this number is not divisible by eight, at least four mercury atoms should occupy the special positions. A possibility of statistical distribution of mercury atoms was omitted, because no anomaly of background intensities was observed on the films. As the space group $C2ca$ has no symmetry element to fix the x coordinate of points, the origin of the cell can be set at one mercury atom of $4a$ and the positions of at least three heavy atoms (one Hg and two Tl atoms) must be found.

The vector set of equipoint $8b$ is illustrated in Fig. 1*b*. These relations of equipoints and the heights of peaks which are expected for the vectors between heavy atoms, limit the probable positions of rotation peaks in the Patterson diagram $P(0,v,w)$, Fig. 1*a*. It was, however, impossible to find uniquely the rotation peaks of heavy atoms and another attempt was made to obtain the heavy-atom positions, because almost all peaks of the Patterson maps have broadened shapes and overlap each other.

If one component atom of a structure occupies the origin of the cell, the Patterson diagram contains the image of the true structure and the origin of the diagram coincides with that of the cell. If this crystal has axial symmetry, the image of the structure, being concentrated in one section, will also be obtained from the Harker section (the implication diagram). Therefore when we superpose the implication diagram on each section of the Patterson diagram in such a way that the origin of the former coincides with one of the symmetry axes in the cell, and when we note the peaks common in both diagrams, the image of the true structure should be found in the resultant map. In the present case, the Patterson section $P(0,v,w)$, which was drawn on a scale of one

² The word "heavy atoms" will be used for mercury and thallium atoms hereafter, because the difference of their atomic scattering factors is quite small.

³ $(0,0,0; \frac{1}{2}, \frac{1}{2}, 0) \div 8b: \quad x, y, z; \quad x, \bar{y}, \bar{z}; \quad x, \frac{1}{2} + y, \frac{1}{2} - z; \quad x, \frac{1}{2} - y, \frac{1}{2} + z.$
 $4a: \quad x, 0, 0; \quad x, \frac{1}{2}, \frac{1}{2}.$

to two, was superposed on $P(0,v,w)$ and $P(\frac{1}{4},v,w)$. The result, however, showed still too many candidates for the heavy atom positions.

In order to eliminate some of the candidates, the relations of the cross vectors between the origin and the points in the general positions were derived. The end points of vectors show special relations as indicated in Fig. 2: these points are related to each other by three twofold axes. One of them designated as I is a component of the symmetry elements in the Patterson space group $Cmmm$, while the others, II and III, being equivalent, are additional ones. When two sets of Patterson diagrams are superposed according to the additional operations, the cross vector peaks between the atom of the origin and the atoms of the general positions can be obtained. This result also includes the image of the actual structure.

The above procedure eliminated most of the peaks in the Patterson section. The implication diagram was also superposed on the resultant maps, and five peaks common in both diagrams were adopted as candidates (Figs. 3*a* and 3*b*). In order to eliminate the number of candidates and to obtain the relative positions of each of the heavy atoms, five sets of minimum functions were evaluated using the origin atom, each of the candidates and their symmetrically equivalent positions. Four minimum-function diagrams drawn with four points I, II, III and IV (Figs. 3*a* and 3*b*) were compatible with each other, and these peaks were considered to correspond to one mercury, two thallium and one antimony atoms. Position I was regarded as the antimony atom, because the peak height in the minimum-function diagram is lower. The relative positions of these peaks, having been impossible to derive by the operation in Fig. 2, are found from minimum-function diagrams. The structure factors were calculated with the coordinates obtained for the reflections which are in the range of $\sin \theta \leq 0.7$. The discrepancy index R was about 43%. The atomic scattering factor of mercury was used for both mercury and thallium atoms at this stage. The remaining atoms were found by successive three-dimensional Fourier and difference Fourier syntheses. In the course of this procedure, the antimony atom and one of three heavy atoms (denoted II in Fig. 3*a*) revealed an anomaly in the difference map; that is, a deep depression was found at the position of the latter and an elevation at the position of the former. Besides, the coordination of sulfur atoms around them was quite strange: atom I, which was regarded as Sb, has two nearest sulfur atoms and the coordination of atom II was a flat trigonal pyramid. Therefore the Sb atom (I) and atom II were

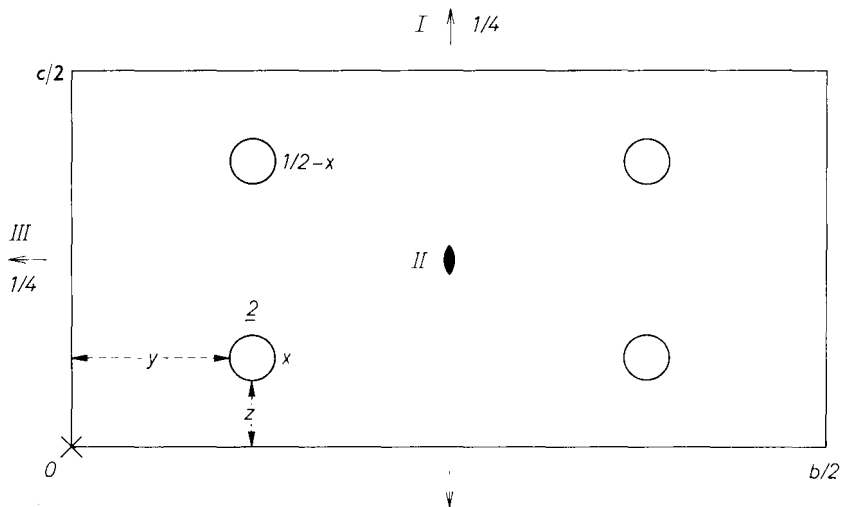


Fig.2. The relations of cross vectors between the origin and the points in the general positions. The underlined values indicate weights of points. I, II and III are directions of the three twofold axes

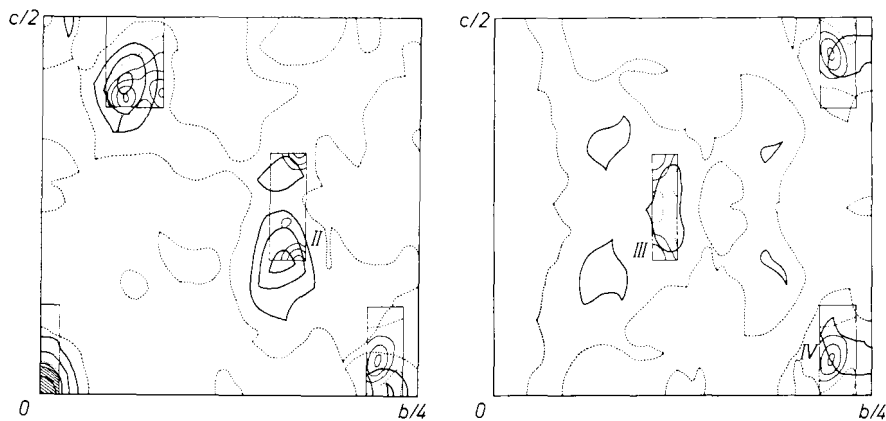


Fig.3. Candidates for heavy atoms. The same diagram as in Fig.1a was used as the implication diagram. All candidates should be in the area within the rectangular sections. (a) $x = 0$. (b) $x = 0.24 \approx \frac{1}{4}$. Independent candidates in this section are only III and IV

interchanged in position. Four arsenic and ten sulfur atoms were found in the further process. The R index was about 22% at this stage and no anomaly was found in the three-dimensional difference synthesis. The mercury atom was distinguished from thallium atoms by the difference in the coordinations around them.

Table 1. Comparison of observed and calculated structure factors for urbaite

b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c
0	2	0	28*	79	0	8	7	123	120	1	21	3	210	216	1	5	12	179	178	2	12	6	307	317
4			358	289	10			118	134	23			275	275	7			150	150	14			67*	57
6			259	267	12			80	88	25			32*	44	9			104	105	16			88	74
8			163	164	14			225	226	27			214	220	11			81	73	18			315	312
10			160	155	16			416	441	1	4		865	943	13			78	78	20			103	99
12			259	243	18			75	73	3			404	418	1	13		86	89	22			298	307
14			271	283	20			122	129	5			283	294	3			252	256	24			39	25
16			244	269	22			18*	4	7			656	711	5			22*	12	2	2	7	586	623
18			1152	1229	24			314	330	9			295	310	7			193	186	4			166	160
20			663	669	0	0	8	297	324	11			538	557	9			147	143	6			335	346
22			93	54	2			268	285	13			278	281	2	0	0	402	170	8			194	210
24			115	97	4			194	214	15			99	81	2			248	210	10			261	266
26			355	352	6			347	371	17			407	400	4			245	228	12			147	148
28			441	459	8			112	118	19			387	387	6			1110	1219	14			343	342
0	2	1	116	32	10			264	297	21			305	300	8			762	795	16			63	61
4			378	376	12			52*	43	23			72	62	10			95*	76	18			146	142
6			450	489	14			55*	25	25			255	262	12			536	539	20			121	112
8			831	918	16			31*	12	27			205	219	14			797	822	22			83	69
10			587	637	18			290	298	1	5		723	764	16			117	129	24			22*	26
12			705	762	20			194	198	3			241	244	18			384	386	2	0	8	263	262
14			174	191	22			45	41	5			247	258	20			695	614	2			421	417
16			285	298	0	2	9	46*	25	7			299	300	22			216	216	4			286	276
18			171	168	4			361	389	9			457	483	24			308	300	6			69	49
20			32*	8	6			264	287	11			123	121	26			198	202	8			529	558
22			131	139	8			292	318	13			235	248	28			78	71	10			474	508
24			105	110	10			180	191	15			169	161	30			573	551	12			38*	35
26			257	256	12			426	451	17			450	467	4	2	1	64*	82	14			165	157
28			14*	10	14			67	66	19			373	378	6			464	483	16			97	91
0	0	2	344	389	16			33*	37	21			260	266	8			49*	53	18			61	20
4			196	193	18			123	123	23			200	200	10			136	100	20			166	169
6			928	1046	20			53	58	25			180	185	12			132	100	22			150	152
8			234	271	0	10		667	699	1	6		241	255	14			464	476	2	2	9	352	349
10			717	781	2			249	254	3			454	476	16			551	556	4			126	139
12			114*	82	4			43*	27	5			351	373	18			113	81	6			239	241
14			152	171	6			200	203	7			239	237	20			382	388	8			120	114
16			453	475	8			186	193	9			166	166	22			89	80	10			184	194
18			93	50	10			61	74	11			339	341	24			316	315	12			201	197
20			175	181	12			65	67	13			147	154	26			131	131	14			97	105
22			500	514	14			56	52	15			445	451	28			182	193	16			240	248
24			457	475	16			96	99	17			249	252	2	0	2	372	372	18			173	166
26			255	255	18			194	201	19			82	86	2			841	816	20			187	189
28			150	149	0	2	11	24*	1	21			193	206	4			273	267	2	0	10	267	255
0	2	3	366	378	4			206	213	23			259	256	6			122*	93	2			422	423
4			1360	1640	6			58	54	25			97	107	8			196	194	4			314	317
6			178	172	8			324	334	1	7		395	417	10			754	780	6			282	295
8			621	678	10			60	57	3			44	46	12			250	255	8			68	64
10			298	322	12			336	352	5			499	528	14			237	206	10			288	287
12			87*	85	14			387	389	7			414	439	16			674	704	12			187	182
14			822	881	0	0	12	161	162	11			124	142	18			257	260	14			301	305
16			489	495	2			146	162	9			296	312	20			254	250	16			212	207
18			20*	9	4			243	239	13			425	432	22			236	234	18			64	62
20			235	238	6			297	297	15			395	399	24			115	110	2	2	11	106	96
22			253	251	8			61	52	17			320	320	26			88	89	4			163	164
24			450	461	10			36	9	19			179	167	28			223	228	6			703	714
26			68	69	12			33	13	21			263	272	2	2	3	369	365	8			173	171
28			27*	8	14			41	15	23			120	119	4			305	303	10			16*	42
0	0	4	242	255	0	2	13	111	111	1	8		207	212	6			199	176	12			286	291
2			24*	59	4			61	59	3			257	268	8			200	204	14			178	179
4			687	766	6			203	216	5			399	428	10			200	206	16			74	78
6			333	388	8			65	65	7			190	195	12			665	684	2	0	12	243	228
8			823	919	10			107	111	9			99	91	14			312	302	4			193	193
10			327	356	1	1	1	204	209	11			148	144	16			138	130	4			101	92
12			672	713	3			98*	136	13			151	164	18			541	548	6			105	109
14			37*	11	5			116*	104	15			174	170	20			313	308	8			54	49
16			214	227	7			388	404	17			128	147	22			73	62	10			193	187
18			173	166	9			835	909	19			155	156	24			326	330	12			153	161
20			238	253	11			701	738	21			185	187	26			248	257	2	2	13	436	427
22			89	80	13			526	543	23			155	158	2	0	4	89	89	4			110	110
24			86	92	15			526	552	1	9		358	365	2			492	514	6			149	149
26			226	240	17			253	257	3			241	242	4			250	231	8			47	51
0	2	5	109	119	19			139	121	5			154	166	6			412	422	3	1	1	171	157
4			276	302	21			260	261	7			193	203	8			113	116	3			237	222
6			32*	68	23			182	188	9			264	281	10			553	567	5			688	688
8			521	577	25			260	254	11			301	312	12			520	534	7			649	649
10			540	583	27			255	254	13			248	256	14			215	206	9			676	676
12			96	88	1	1	2	141	143	15			302	302	16			62*	49	11			551	536
14																								

Table 1. (Continued)

b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c	b	k	l	F _o	F _c
3	5	5	141	125	3	9	11	242	249	4	10	6	439	434	5	9	5	445	428	5	9	12	38*	45
7			206	199	11			178	170	12			353	351	11			389	374	11			101	97
9			492	479	13			158	157	14			59*	52	13			118	134	5	1	13	98	97
11			488	487	15			158	150	16			96	91	15			386	370	3			95	88
13			125	116	3	1	12	150	139	18			277	272	17			299	275	6	0	0	111*	98
15			410	398	3			87	92	20			233	237	19			114	90	2			347	302
17			368	356	5			49	48	22			120	123	21			272	267	4			220	203
19			39*	34	7			52	43	24			100	99	23			208	196	6			940	910
21			291	291	9			34	43	4	2	7	210	221	25			99	92	8			738	707
23			283	271	11			89	74	4			420	411	5	1		524	512	10			76*	63
25			82	76	13			80	73	6			155	144	3			340	341	12			346	318
27			162	159	3	1	13	61	63	8			156	148	5			278	255	14			734	699
3	1	4	617	624	3			89	93	10			164	171	7			578	555	16			244	225
5			494	504	5			21*	15	12			363	351	9			153	153	18			415	390
7			260	247	7			263	254	14			47*	28	11			543	520	20			570	552
9			591	583	4	0	0	2102	2420	16			701	706	13			180	160	22			191	190
11			160	151	2			118*	103	18			143	138	15			169	165	24			263	255
13			592	586	4			72*	66	20			81	75	17			298	283	26	2	1	415	392
15			210	169	6			380	364	22			61	62	19			349	338	6	2	1	415	392
17			117	102	8			308	299	4	8		150	145	21			235	232	4			161	114
19			418	402	10			320	313	2			260	251	23			66	61	6			435	414
21			297	291	12			1065	1039	4			287	283	25			270	274	8			133	121
23			293	286	14			150	154	6			402	403	5	1	5	573	544	10			111	55
25			84	79	16			409	404	8			235	218	3			322	321	12			39*	47
27			246	248	18			907	895	10			313	312	5			414	418	14			364	349
3	1	5	98	104	20			467	438	12			62	42	7			301	294	16			457	427
5			739	730	22			233	230	14			152	147	9			427	421	18			82	65
7			281	284	24			106	99	16			285	287	11			125	120	20			373	365
9			486	484	26			147	145	18			141	133	13			235	225	22			28*	40
11			346	350	4	2	1	154	149	20			208	205	15			316	299	24			25*	241
13			654	659	4			535	540	22			155	155	17			342	335	26			163	167
15			162	159	6			308	288	4	2	9	133	130	19			298	298	6	0	2	303	309
17			312	310	8			1023	1009	4			251	250	21			184	182	2			743	718
19			217	222	10			751	727	6			194	195	23			271	271	4			137	135
21			381	384	12			428	416	8			379	384	25			188	186	6			157	152
23			365	368	14			51*	29	10			341	328	5	1	6	208	216	8			233	215
25			295	300	16			0*	29	12			372	368	3			581	567	10			715	681
27			254	255	18			288	279	14			53	37	5			398	394	12			260	247
3	1	6	179	178	20			196	189	16			6*	33	7			203	188	14			185	156
5			251	254	22			34*	32	18			222	217	9			146	134	16			702	672
7			601	600	24			267	263	20			185	180	11			303	295	18			251	231
9			482	489	26			197	197	4	0	10	492	469	13			169	156	20			227	219
11			128	124	4	0	2	183	177	2			252	239	15			419	411	22			160	167
13			159	167	2			205	187	4			84	82	17			187	182	24			64	63
15			188	204	4			1626	1778	6			250	245	8			259	245	26			85	87
17			136	126	6			214	209	8			258	239	21			163	163	6	2	3	373	351
19			346	337	8			249	244	10			74	69	23			264	262	4			260	258
21			262	262	10			50*	30	12			182	174	5	1	7	324	325	6			222	172
23			104*	17	12			104	99	14			48	49	3			483	473	8			197	190
25			309	300	14			603	593	16			150	145	5			378	358	10			246	235
3	1	7	326	327	16			513	489	18			184	190	7			393	377	12			651	611
5			35	23	18			191	182	4	2	11	134	120	9			91	80	14			219	208
7			419	402	20			164	162	4			70	61	11			188	194	16			153	141
9			468	476	22			606	596	6			79	71	13			350	345	18			557	517
11			359	348	24			128	121	8			139	137	15			489	488	20			319	296
13			466	462	26			51	42	10			41	40	17			259	251	22			27*	31
15			100	92	4	2	3	273	261	12			291	292	19			254	253	24			289	287
17			221	217	4			1506	1565	14			297	290	21			276	277	6	0	4	70*	75
19			444	443	6			120	105	4	0	12	30*	21	23			157	161	2			458	453
21			452	451	8			465	427	2			200	189	5	1	8	180	189	4			28*	275
23			296	291	10			211	208	4			170	160	3			135	118	6			294	284
25			224	231	12			178	115	6			42	32	5			253	243	8			168	159
3	1	8	301	305	14			888	830	8			135	129	7			197	189	10			450	430
5			124	127	16			487	472	10			66	60	9			11*	35	12			448	420
7			241	222	18			85*	64	12			124	116	11			276	261	14			201	181
9			199	196	20			94	96	4	2	13	50	47	13			250	238	16			68*	78
11			372	368	22			281	270	4			367	353	15			174	169	18			138	131
13			241	248	24			441	439	6			52	47	17			16*	29	20			140	134
15			51*	54	26			88	85	5	1		156	143	19			157	153	22			394	385
17			205	199	4	0	4	225	206	3			82*	100	21			100	103	24			65	69
19			312	312	2			100*	101	5			596	586	5	1	9	395	379	6	2	5	223	214
21			214	220	4			236	225	7			628	601	3			140	130	4			64*	41
23			69	51	6			408	397	9			544	523	5			207	210	6			518	520
25			162	156	8			1022	1033	11			395	381	7			79	82	8			236	226
3	1	9	104	96	10			363	335	13			326	495	9			336	330	10			172	150
5			367	355	12			408	396	15			149	137	11			127						

Table 1. (Continued)

h	k	l	$ F_o $	$ F_c $	h	k	l	$ F_o $	$ F_c $	h	k	l	$ F_o $	$ F_c $	h	k	l	$ F_o $	$ F_c $					
6	14	7	441	430	7	9	5	547	521	8	4	3	1147	1142	9	11	2	257	231	10	14	1	376	355
16	93	75						95	78				116	112	13			128	128	16			420	405
18	112	111						137	192				543	506	15			200	184	18			109	118
20	133	128						187	188				276	257	17			97	107	20			205	205
22	68	65						429	412				90	105	19			123	134	22			84	87
6	0	8	264	252				365	371				701	650	21			185	185	10	0	2	351	352
2			393	386				274	278				408	382	23			104	110				507	510
4			334	322				142	145				52*	4	9	1	3	302	300	4			224	227
6			81	75	7	1	6	220	229				197	186	3			338	333	6			67*	78
8			539	534				424	424				197	197	5			127	102	8			65*	42
10			453	445				228	223				577	381	7			132	122	10			459	439
12			45*	33				170	158			8	0	4			115	115	9			446	410	
14			159	152				79	37				71*	66	11			361	327	14			124	109
16			65	47				214	201				465	459	13			72	73	16			437	427
18			26*	24				91	90				313	294	15			164	161	18			238	227
20			152	148				289	279				721	691	17			167	158	20			207	205
6	2	9	449	406				247	240				158	148	19			175	173	22			223	234
4			172	168				77	74				557	525	21			173	173	10	2	3	162	168
6			250	240				217	225				27*	27	23			224	228	4			220	219
8			121	127	7	1	7	324	329				176	171	9	1	4	447	446	6			125	87
10			89	89				389	386				167	154	3			197	194	8			108	98
12			159	152				433	427				213	218	5			197	192	10			128	127
14			102	106				326	317				57	59	7			408	484	12			502	474
16			214	211				209	192	8	2	5	86	81	9			155	140	14			235	228
18			80	73				275	261				260	255	11			431	416	16			148	140
6	0	10	229	214				450	437				84	48	13			161	155	18			344	341
2			387	371				369	362				419	406	15			188	180	20			212	211
4			317	300				343	337				422	416	17			195	190	10	0	4	133	135
6			279	273				109	113				90	72	19			271	275	2			328	331
8			60	56				208	209				285	272	21			197	202	4			114	104
10			256	254	7	1	8	201	184				355*	75	9	1	5	483	500	6			327	314
12			146	135				265	263				158	160	11			175	172	8			120	104
14			271	267				471	462				305	305	5			133	136	10			370	352
16			173	169				32*	18				27*	28	7			273	272	12			430	408
6	2	11	61	64				36*	30	8	0	6	496	511	9			245	236	14			141	126
4			159	147				48*	11				164	161	11			251	250	16			10*	33
6			615	599				231	229				62*	51	13			187	176	18			165	161
8			161	156				181	174				450	437	15			142	136	20			101	98
10			97	94				179	174				630	614	17			292	295	10	2	5	271	283
12			237	228				195	186				477	455	19			236	247	4			117	114
6	0	12	165	159				200	197				722	697	21			153	155	6			337	335
2			185	176	7	1	9	132	137				61	80	9	3	6	188	190	8			150	137
4			79	72				174	184				92	98	3			284	299	10			85	54
6			70	70				255	250				109	96	5			301	307	12			277	275
8			29*	36				225	217				79	82	7			161	150	14			157	158
7	1	1	288	292				334	319	8	2	7	46*	51	9			46*	48	16			217	230
3			123	86				259	255				553	553	11			274	273	18			74	77
5			435	402				251	247				88	69	13			115	110	20			146	151
7			857	823				202	196				152	110	15			375	375	10	0	6	137	142
9			673	630	7	1	10	74	69				128	129	17			143	142	2			0*	23
11			515	468				245	250				74	77	19			41	38	4			245	264
13			495	436				86	81				156	154	3			336	346	6			58*	43
15			129	116				171	179				351	332	9	3	7	312	323	8			461	454
17			176	166				49	51				59	65	5			311	318	10			55*	15
19			228	210				267	260				107	106	7			331	329	12			310	309
21			214	207				32	34	8	0	8	245	253	9			108	98	14			51	35
23			175	179	7	1	11	80	88				222	218	11			209	209	16			100	105
25			226	234				129	138				161	162	13			239	255	18			243	243
7	1	2	43*	73				109	119				319	310	15			290	282	10	2	7	473	488
3			485	488				176	178				88	76	17			212	208	4			157	156
5			631	608				218	217				248	246	9	1	8	97	100	6			226	235
7			267	252	8	0	0	2521	2789				38*	40	3			138	138	8			173	176
9			66*	26				122	97				47*	53	5			220	222	10			178	167
11			114	101				323	306				24*	22	7			230	228	12			124	122
13			468	435				163	148				239	230	9			121	128	14			125	121
15			300	272				81*	76	8	2	9	83	85	11			215	212	16			32	30
17			168	156				157	135				308	310	13			96	81	10	0	8	165	161
19			168	157				328	305				231	219	15			117	114	2			263	266
21			255	248				276	258				252	254	9	1	9	345	354	4			155	151
23			181	193				348	334				148	3	3			232	235	6			87	84
25			236	248				975	938				37	359	5			111	113	8			354	342
7	1	3	339	322				355	337				47	36	7			74	82	10			356	361
5			425	419				74	83	8	0	10	514	541	9			198	194	12			58	52
5			283	277				86	83				204	198	11			169	168	14			128	129
7			180	169	8	2	1	315	128				49	51	13			144	143	10	2	9	152	155
9			656	586				320	310				170	174	9	1	10	74	72	4			144	140
11			173	169				315	297				161	158	3			223	226	6			119	114
13			230	207				695	625				84	79	5			134	153	8			45	45
15			230	222				467	433				58											

Table 1. (Continued)

h	k	l	F _o	F _c	h	k	l	F _o	F _c	h	k	l	F _o	F _c	h	k	l	F _o	F _c							
11	11	2	125	102	11	9	8	40	38	12	4	5	225	239	13	5	4	214	223							
13	400	380	11	61	59	6	49*	21	7	290	291	2	269	271	14	0	4	114	125							
15	254	218	11	1	9	186	196	8	133	143	9	111	106	4	185	192	4	185	192							
17	147	148	3	93	98	10	222	220	11	246	247	6	125	127	6	125	127	6	125	127						
19	96	90	5	110	118	12	34*	31	13	113	112	8	172	172	8	172	172	8	172	172						
11	3	5	268	205	7	188	148	14	219	224	15	118	114	10	148	152	10	148	152							
3	341	333	9	222	227	16	36	17	13	1	5	220	225	12	221	229	12	221	229							
5	100	82	12	0	0	1354	1337	12	0	6	729	766	3	200	207	14	2	5	57	59						
7	189	181	2	112	106	2	97	107	5	166	181	4	98	99	4	98	99	4	98	99						
9	272	261	4	110	105	4	134	136	7	219	224	6	303	318	6	303	318	6	303	318						
11	363	334	6	348	347	6	269	280	9	79	86	8	138	148	8	138	148	8	138	148						
13	61*	50	8	221	215	8	280	285	11	97	90	10	118	132	10	118	132	10	118	132						
15	278	253	10	226	215	10	275	278	13	116	120	14	0	6	145	147	14	0	6	145	147					
17	237	237	12	583	568	12	271	284	13	1	6	173	180	2	100	97	2	100	97	2	100	97				
19	80	78	14	97	110	14	54	49	3	274	289	4	123	124	4	123	124	4	123	124						
11	1	4	375	379	16	221	219	12	2	7	102	101	5	167	177	6	48	35	6	48	35					
3	352	362	18	511	516	4	267	271	7	163	170	8	277	291	8	277	291	8	277	291						
5	233	229	12	2	1	124	132	6	96	92	9	60	58	14	2	7	156	157	14	2	7	156	157			
7	315	301	4	344	337	8	79	81	11	212	225	15	1	1	233	233	15	1	1	233	233					
9	115	99	6	137	138	10	100	101	13	1	7	211	221	3	64	60	3	64	60	3	64	60				
11	359	339	8	535	520	12	211	220	13	3	7	268	280	5	245	244	5	245	244	5	245	244				
13	210	205	10	379	367	12	0	8	66	56	5	225	233	7	466	471	7	466	471	7	466	471				
15	100	88	12	216	213	2	161	166	7	149	155	9	382	388	9	382	388	9	382	388						
17	309	317	14	62	68	4	157	159	9	103	107	11	296	300	11	296	300	11	296	300						
19	138	140	16	62	41	6	256	261	13	8	77	75	15	1	2	78	76	15	1	2	78	76				
11	5	5	333	355	18	166	171	8	178	181	3	53	53	3	280	288	3	280	288	3	280	288				
3	204	200	12	0	2	315	316	12	2	9	94	93	14	0	0	120	88	5	371	376	5	371	376			
5	266	266	2	147	140	13	1	1	104	98	2	264	251	7	195	198	7	195	198	7	195	198				
7	303	312	4	843	855	3	95	79	4	159	159	9	34*	26	9	34*	26	9	34*	26	9	34*	26			
9	479	482	6	102	89	5	281	272	6	429	419	11	115	116	11	115	116	11	115	116	11	115	116			
11	146	148	8	154	163	7	328	329	8	399	391	15	1	3	155	150	15	1	3	155	150	15	1	3	155	150
13	234	236	10	42*	40	9	336	323	10	122	117	3	282	287	3	282	287	3	282	287	3	282	287			
15	88	97	12	153	153	11	174	175	12	50	44	5	144	149	5	144	149	5	144	149	5	144	149			
17	294	303	14	364	358	13	244	235	14	154	370	7	116	124	7	116	124	7	116	124	7	116	124			
19	195	195	16	285	291	15	131	134	1	140	139	9	281	292	9	281	292	9	281	292	9	281	292			
11	1	6	340	354	18	164	167	17	80	79	4	150	151	15	1	4	402	422	15	1	4	402	422			
3	307	324	12	2	3	168	165	13	1	2	123	116	6	178	176	3	346	359	6	178	176	3	346	359		
5	145	145	4	828	838	3	173	166	8	148	149	5	144	145	5	144	145	5	144	145	5	144	145			
7	100	95	6	61*	55	5	215	211	10	17*	22	7	227	236	7	227	236	7	227	236	7	227	236			
9	45*	42	8	334	315	7	72	89	12	33*	24	15	330	346	15	330	346	15	330	346	15	330	346			
11	52	51	10	171	161	9	108	113	14	143	139	3	195	199	3	195	199	3	195	199	3	195	199			
13	161	163	12	164	158	11	196	197	14	0	2	227	239	5	148	152	5	148	152	5	148	152				
15	210	212	14	475	465	13	106	113	2	376	368	16	0	0	1141	1169	16	0	0	1141	1169	16	0	0	1141	1169
17	343	347	16	304	310	15	125	133	4	112	126	2	91	98	2	91	98	2	91	98	2	91	98			
19	260	270	18	26	26	17	68	4	6	71	68	4	282	285	4	282	285	4	282	285	4	282	285			
11	1	7	210	216	12	0	4	103	82	13	3	70	79	8	132	128	8	132	128	8	132	128				
3	290	301	2	135	134	3	325	320	10	346	345	16	2	1	100	97	16	2	1	100	97	16	2	1	100	97
5	93	91	4	135	140	5	199	199	12	108	113	4	202	215	4	202	215	4	202	215	4	202	215			
7	157	155	6	276	271	7	226	220	14	34	36	6	103	103	6	103	103	6	103	103	6	103	103			
9	292	304	8	603	603	9	234	225	14	2	3	164	164	16	0	2	146	141	16	0	2	146	141			
11	247	251	10	243	242	11	174	176	4	186	179	2	102	100	2	102	100	2	102	100	2	102	100			
13	114	115	12	310	298	13	112	113	6	105	97	4	362	373	4	362	373	4	362	373	4	362	373			
15	186	186	14	56	59	15	213	219	8	91	91	8	91	91	8	91	91	8	91	91	8	91	91			
17	301	304	16	79	71	13	1	4	240	240	10	171	171	10	171	171	10	171	171	10	171	171				
19	188	192	12	2	5	52*	28	3	115	118	12	358	375	12	358	375	12	358	375	12	358	375				

* unobserved reflection

Several cycles of least-squares refinement in which individual form factors were used for Hg and Tl atoms and in which isotropic temperature factors for each atom were varied, reduced *R* to 11.0%. During these calculations, an equal weight was used for all reflections. Non-ionized atomic form factors given by IBERS, THOMAS *et al.*, THOMAS and UMEDA, FREEMAN and WATSON, and DAWSON were employed for Hg, Tl, Sb, As and S respectively (*International tables*, 1962). Additional cycles of least-squares refinement in which anisotropic temperature factors were varied, reduced *R* to 4.8% for all 1772 reflections. In the course of these calculations, individual weights calculated by the modified formula of GABE (1966) was used for each of the reflections:

$$w = \frac{1}{\sigma^2(F)} = 4F_0^2 \left/ \sum_{i=1}^4 \left\{ \left(\frac{\partial F_0^2}{\partial q_i} \right)^2 \sigma^2(q_i) \right\} \right.,$$

where *q*₁ = peak count, *q*₂ = background count, *q*₃ = (*LP*)⁻¹ and *q*₄ = transmission. At the final stage, the effect of anomalous dis-

