

EXAFS and XRD investigations of zeunerite and meta-zeunerite

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Abstract. In this paper EXAFS was used to determine bond lengths in the structures of zeunerite and meta-zeunerite. The atomic distances between heavy and light scatterers observed using EXAFS in meta-zeunerite deviate approximately 0.1 Å from literature data of single-crystal X-ray diffraction measurements. Because this difference is significant higher than the error limits of EXAFS measurements, the complete crystal structure of meta-zeunerite, $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 8 \text{H}_2\text{O}$, is revised by X-ray structure analysis. The bond length determinations by EXAFS and the revised XRD data agree within the experimental error limits. In this study EXAFS spectroscopy has proven to be an useful tool for determining precise local bond lengths in the environment of heavy atoms. Moreover, the crystal structure of zeunerite, $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$, hitherto not been described in the literature, was investigated. Reflex broadening effects and intergrowth relationship between zeunerite and meta-zeunerite show that meta-zeunerite grows in nature due to dehydration of zeunerite. The structural transition from zeunerite to meta-zeunerite is connected with a change in the uranyl arsenate layer arrangement and the crystal water content.

1. Introduction

X-ray diffraction (XRD) techniques allow a precise analysis of crystal structures. With modern equipment, even hydrogen atoms can be determined adjacent to heavy atoms. However, in the presence of very heavy atoms, crystal structure determination by single-crystal X-ray diffraction measurements is sometimes less accurate. The reflection intensities are mainly influenced by the heavy scatterers and the error in atomic coordinates increases for the light atoms. The difficulties increase if only crystals of poor quality are available or problems in symmetry determination appear. In these cases, a bond length determination independent from crystal quality and the knowledge of lattice parameters is helpful. Extended X-ray Absorption Fine Structure (EXAFS) spectroscopy allows determining the distances between heavy atoms and the immediately surrounding atoms. The estimated standard deviations for

bond lengths are less than 0.02 Å; the accuracy decreases for more distant shells. EXAFS gives only the average bond length in each single coordination shell. As shown in this paper, the atomic distances determined by EXAFS on natural meta-zeunerite deviate considerably from the single crystal structure data of Hanic [1]. Similar differences between EXAFS and XRD measurements for other uranium compounds are described in the literature [2]. For this reason and taking into account that Hanic [1] could use only intensities from Weissenberg photographs for his structure analysis, a complete redetermination of the meta-zeunerite structure was carried out. EXAFS was used furthermore to distinguish structural differences between zeunerite and meta-zeunerite. The crystal structure of zeunerite is described here for the first time.

Zeunerite and meta-zeunerite are secondary grown minerals, which arise from uranium and arsenic primary minerals in the oxidized zone of rock deposits. They belong to the structure family with the chemical formula $\text{A}^{m+}[\text{UO}_2\text{XO}_4]_m \cdot n \text{H}_2\text{O}$, where XO_4 appears as phosphate or arsenate and A is a hydrated monovalent or divalent cation [3, 4]. In nature, both A and X occur often mixed by isomorphous replacement. Because of the wide range of the interlayer cations A^{m+} , this group includes a lot of structurally related members [5]. The main structural principles are well known [3] and will be summarized here shortly. Each $[\text{UO}_2]^{2+}$ unit is built up by uranium with two double bonded oxygen atoms in axial direction (O_{ax}). This uranyl unit is surrounded in the equatorial plane by four oxygen atoms (O_{eq}) in a square planar arrangement. Tetrahedra of $[\text{XO}_4]^{3-}$ ions and tetragonal dipyramidal coordinated uranyl ions $[\text{UO}_2]^{2+}$ build stable layers. These $[\text{UO}_2\text{XO}_4]_{\infty}$ layers are connected forming crystals with a tetragonal or pseudotetragonal morphology and a platy (001) habit. The uranyl arsenate layer structure causes perfect cleavage. Charge neutrality of the uranyl arsenate layers is given by hydrated $[\text{A}(\text{H}_2\text{O})_4]^{m+}$ cations.

Zeunerite and meta-zeunerite are two different hydrates of copper uranyl arsenate. The interlayer water content varies depending on vapor pressure and temperature. Three stable hydrate species of $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot n \text{H}_2\text{O}$ have been described [6, 7]: (i) the completely hydrated phase, zeunerite, with $n = 12 \text{H}_2\text{O}$, (ii) meta-I hydrate containing $n = 8 \text{H}_2\text{O}$, naturally occurring as meta-zeunerite and (iii) meta-II hydrate containing $n = 2-5 \text{H}_2\text{O}$.

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Meta-II hydrate is only recognized under laboratory conditions but does not occur naturally. There are more dehydration stages proposed in the meta-II hydrate region [8]. The transition between the completely hydrated phase and the meta-I hydrate occurs near ambient temperatures and at low humidity during a long time. At 55 °C the transition from zeunerite to meta-zeunerite happens within two hours [7]. Problems concerning the determination of the crystal symmetry and crystal structure may occur, especially if two hydrate species co-exist. According to the tetragonal crystal symmetry, zeunerite and meta-zeunerite should be optically uniaxial. Weak deviations from expected uniaxial behavior were described for rare cases. Deviations from uniaxial optic are very small for meta-zeunerite. An optical axis angle of maximal $2V = 8^\circ$ for zeunerite is mentioned [7], which probably arises from a hydrate transition state. But this has not been confirmed by X-ray diffraction methods.

In a first step, the EXAFS results estimated on zeunerite/meta-zeunerite will be represented. Deviations of the obtained bond lengths from X-ray diffraction literature data [1] will be discussed. In a second step, the single crystal structure analysis on zeunerite and meta-zeunerite will be described and compared with the EXAFS results.

2. Experimental

2.1. Samples

Natural zeunerite/meta-zeunerite often contains phosphate as isomorphous replaced element instead of arsenate. Therefore, two samples from a collection of different zeunerite minerals were selected with a negligible phosphorus content ascertained by microprobe analysis. The phase content was analyzed by powder X-ray diffraction. An intergrown zeunerite/meta-zeunerite mineral from Wheal Basset, Cornwall/England (sample 1) was used for EXAFS, powder and single-crystal X-ray diffraction measurements. It should be noted that the zeunerite content in this sample can be ignored for EXAFS measurements. A pure zeunerite crystal fragment was separated for XRD measurements from the same sample. An additional sample of meta-zeunerite from the mine "Weißer Hirsch", Schneeberg/Saxony in Germany, (sample 2) consisting of very small crystals was used for single-crystal X-ray diffraction measurements. Furthermore, an attempt was made to obtain single crystals by synthesis according to the procedure described in the literature [8]. The resulting compound (sample 3) has been proven by powder X-ray diffraction to be identical with zeunerite, $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$. Because no single crystals were obtained by synthesis, sample 3 was used only for low-temperature Cu K-edge EXAFS measurements.

2.2. EXAFS measurements

EXAFS measurements were carried out on the Rossendorf Beamline (ROBL) [9] at the European Synchrotron Radiation Facility (ESRF) under dedicated ring conditions (6.0 GeV, 100–200 mA). The monochromator, which is

equipped with a water-cooled Si(111) double-crystal system, was used in channel-cut measuring mode. Higher harmonics were rejected by two Pt coated mirrors. The first mirror collimates the X-ray beam onto the monochromator crystal. The second mirror focuses the beam vertically to the sample. The vertical slit aperture before the sample was set to 0.8 mm. A monochromator feedback control system [10] was used for suppressing the decay of the primary X-ray flux. Energy steps were calculated giving corresponding k -space steps of 0.05 \AA^{-1} . The sample was sealed with polyethylene foil for safety reasons. All measurements were taken at ambient conditions with the exception of two spectra measured at the Cu–K edge. Samples 1 and 3 were ground in an agate mortar, mixed with boron nitride and pressed as 1.3 cm diameter pellet. The amount of uranium used was calculated to give a jump of one across the uranium L_{III} -edge. The same sample was used for the As K-edge and for the Cu K-edge measurements. Uranium L_{III} -edge and arsenic K-edge EXAFS were collected in transmission geometry using argon-filled ionization chambers. Two scans were recorded for each energy range, and the spectra were averaged. The copper K-edge EXAFS spectrum at room temperature was measured in fluorescence mode using a four-pixel Ge detector [11]. In order to distinguish structural differences between $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$ and $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 8 \text{H}_2\text{O}$ by reducing thermal oscillations through cooling the samples to 15 K, a closed-cycle He cryostat was used. Eight single scans were averaged for the Cu K-edge fluorescence spectrum measured at $T = 298 \text{ K}$. For the measurements at $T = 15 \text{ K}$, only two scans were used. Sample orientation during the fluorescence measurements was 45° to the beam. Because of the polarization dependence, transmission measurements were carried out using a sample with a normal orientation of 0° and 45° to the beam direction. Metal foils were used to provide energy calibration references. First inflection points at the Zr K-edge at 17995.9 eV and at the Au L_{III} -edge at 11919.7 eV [12] were used for energy calibration. EXAFS data was extracted from the raw absorption spectra by standard methods using the computer program EXAFSPAK [13].

2.3. X-ray diffraction measurements

In order to ascertain the relation of zeunerite and meta-zeunerite in sample 1, a quantitative phase analysis from X-ray powder diffraction measurement was carried out with the program PowderCell [14]. For this measurement a quartz capillary with a diameter of 0.3 mm was filled with a fine powder of sample 1. The powder pattern was recorded on a STOE StadiP transmission diffractometer in Debye-Scherrer geometry using CuK_α radiation ($\lambda = 1.5405 \text{ \AA}$) in a 2θ range of 5° to 100° with a step width of 0.02° . The intergrowth relationship between zeunerite and meta-zeunerite in sample 1 was identified by Weissenberg photographs. This measurement was performed using unfiltered Cu radiation. Intensity data for single-crystal analyses were collected on a Siemens SMART three-circle diffractometer equipped with a CCD area detector. Details of data collection for the crystal structure determination of zeunerite and meta-zeunerite are listed in Table 1. The first

Table 1. Details of data collection, structure determination and crystallographic data of zeunerite and meta-zeunerite. Estimated standard deviations are given in parentheses.

Mineral	Zeunerite	Meta-zeunerite (ordered)	Meta-zeunerite (disordered)
Formula	Cu[UO_2AsO_4] ₂ · 12 H ₂ O	Cu[UO_2AsO_4] ₂ · 8 H ₂ O	
Crystal system	tetragonal	tetragonal	
Space group	<i>I4/mmm</i>	<i>P4/ncc</i>	<i>P4/nmm</i>
	(No. 139)	(No. 130 2)	(No. 129 2)
Lattice constants a [Å]	7.1751(3)	7.1065(6)	7.1065(6)
c [Å]	20.8728(12)	17.4195(11)	8.7095(11)
V [Å ³]	1074.57(9)	879.73(12)	439.85(8)
Z	2	2	1
<i>d</i> _{calculated} [g · cm ⁻³]	3.392	3.872	3.872
Absorption μ [mm ⁻¹]	19.173		23.390
<i>F</i> (000)	990	910	455
Crystal size [mm ³]	0.10 × 0.05 × 0.02		0.10 × 0.07 × 0.02
Diffractometer	Siemens SMART with CCD area detector		
Radiation/ λ [Å]	MoK α /0.71073		
Temperature [K]	293		
Scan mode	ω		
2 θ range [°]	8.85–7.6	7.45–7.4	
Index ranges	–9 ≤ <i>h</i> ≤ 9 –9 ≤ <i>k</i> ≤ 9 –15 ≤ <i>l</i> ≤ 28	–8 ≤ <i>h</i> ≤ 9 –9 ≤ <i>k</i> ≤ 9 –23 ≤ <i>l</i> ≤ 13	–8 ≤ <i>h</i> ≤ 9 –9 ≤ <i>k</i> ≤ 9 –11 ≤ <i>l</i> ≤ 6
No. of reflections measured	3104	3638	2352
unique	400	506	352
Absorption correction		semi-empirical ψ -scan	
Structure solutions		SHELX-97	
Parameters refined	27	29	34
<i>R</i> 1 (on <i>F</i>)	0.037	0.056	0.029
<i>wR</i> 2 (on <i>F</i> ²)	0.085	0.132	0.072
Goodness of Fit	1.04	0.87	0.87
ΔQ_{max} [e · Å ⁻³]	1.11	0.99	0.96
ΔQ_{min} [e · Å ⁻³]	–1.63	–1.02	–1.82

crystal piece investigated from sample **1** contained zeunerite and meta-zeunerite so that the *hk0* reflections of both phases coincided completely. Some *hkl* reflections are partly overlapped. Nevertheless, both crystal structures could be identified. In order to achieve a better comparison of bond lengths obtained by X-ray diffraction and EXAFS measurements, a piece consisting exclusively of zeunerite was separated from the center of a large crystal from sample **1**. Because of the better crystal quality, the results for meta-zeunerite obtained from sample **2** are presented here. Both structures were solved by direct methods and difference Fourier syntheses. Atomic coordinates and anisotropic displacement parameters were refined by full-matrix least-squares calculations¹.

3. Results and discussions

3.1. EXAFS investigations

To find a set of characteristic bond lengths, the EXAFS spectra were measured at the U L_{III}-edge, at the As K-

edge and at the Cu K-edge. Fourier transforms (FT) of the EXAFS signals represent radial distribution functions of the near-neighbors around the absorber atom. The FT peaks appear at lower R values relative to the true near-neighbor distances as an effect of the EXAFS phase shift. This phase shift influences the bond length determination by additional terms Δ , which are different for each neighboring atom. The program FEFF8 [15] was used to calculate theoretical backscattering phase and amplitude functions. This theoretical approach describes the photoelectron final states by an *ab initio* curved-wave multiple-scattering calculation using an energy-dependent muffin-tin potential. Final-state potentials were calculated using atomic clusters derived from the known atomic coordinates of uranyl arsenate structures as referred in [16]. The phase and amplitude functions were determined by U L_{III}-edge scattering potentials approximated with 52 atoms in a 16-shell cluster with a radius of 5.4 Å. Phase and amplitude functions for the scattering pairs As–O_{eq} and As–U were determined using a cluster with a radius of 4.6 Å. In solid samples the coordination numbers are influenced by polarization effects due to the preferred orientation of the crystallites. The preferred orientation is caused by uniaxial pressure during sample preparation as described in [17]. The atomic distances can be also slightly influenced by the effects of polarization [18]. In order to investigate the influence of polarization effects on the reliability of the determined values, the transmission experiments were performed at two different sample orientations.

¹ Additional material to this paper can be ordered referring to the no. CSD 412820 (As₂CuH₂₄O₂₄U₂), CSD 412821 (As₂CuH₁₆O₂₀U₂/m-zeu130), and CSD 412822 (As₂CuH₁₆O₂₀U₂/m-zeu129), names of the authors and citation of the paper at the Fachinformationszentrum Karlsruhe, Gesellschaft für wissenschaftlich-technische Information mbH, D-76344 Eggenstein-Leopoldshafen, Germany. The list of *F*/*F*_c-data is available from the author up to one year after the publication has appeared.

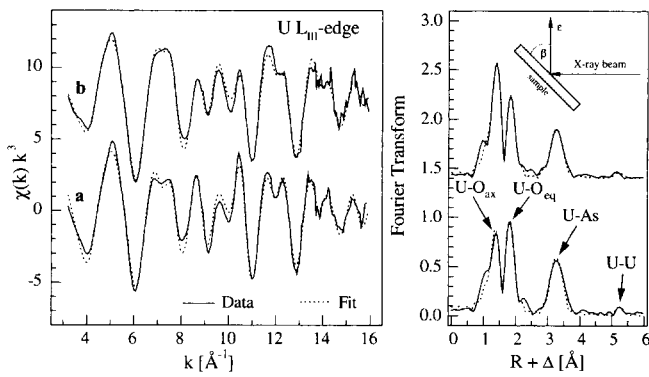


Fig. 1. U L_{III}-edge k^3 -weighted EXAFS spectra of copper uranyl arsenate sample **1** (left) and the corresponding Fourier transforms (right) with $\beta = 0^\circ$ (a) and $\beta = 45^\circ$ (b) measured at $T = 298$ K. The angle β is defined as tilt angle between the sample surface and the polarization vector ϵ .

3.1.1. Uranium L_{III}-edge EXAFS

Uranium k^3 -weighted polarization dependent EXAFS spectra of sample **1** are shown in Fig. 1. Measurements were carried out using two sample orientations, given by the tilt angle β . This angle is defined as the angle between the polarization vector ϵ and the sample surface (see Fig. 1). Structural parameters obtained from the fit procedures are given in Table 2. FT features in the figures are uncorrected for phase shifts. In the U L_{III}-edge FT, the first shell represents the axial oxygen atoms, O_{ax}, at a distance of 1.77–1.79 Å. The smallest resolvable difference in interatomic distances for two different bond lengths in EXAFS measurements is given by the ratio $\Delta R = \pi/(2 \Delta k)$, where Δk is the k -range of the fitted data. For the k -range of 13 Å⁻¹ used for the U L_{III}-edge spectra, the resolution ΔR is 0.12 Å. However, the expected U–O_{ax} bond length differences are not sufficiently large that they could be refined in the least-square fit without constraining some fit parameters. The average U–O_{ax} bond distance given by EXAFS is around 0.1 Å shorter than the averaged value of the axial bond distances 1.94 Å and 1.78 Å reported in the literature [1]. The sec-

ond shell in the FT corresponds to the bond distance of four symmetry-equivalent equatorial atoms (O_{eq}) with a bond length of 2.29 Å. For comparison, the corresponding value given in [1] is 2.18 Å. The coordination numbers for the axial oxygen atoms, N_{O_{ax}}, and for the equatorial oxygen atoms, N_{O_{eq}}, deviate significantly from the expected values. This is due to polarization effects in the EXAFS signal and caused by the preferred orientation of the uranyl arsenate layers with respect to the X-ray beam direction. For a tilt angle $\beta = 0^\circ$ the coordination number N_{O_{ax}} is 1.6 and N_{O_{eq}} is 4.5 atoms. N_{O_{ax}} is lower and N_{O_{eq}} is higher than the crystallographic value. At a tilt angle $\beta = 45^\circ$ the relation is reversed. The coordination number N_{O_{ax}} is 2.2 and N_{O_{eq}} is 3.2. Scattering from the arsenic atoms contributes to a significant third FT peak with a calculated U–As distance of 3.69–3.70 Å. The U–U scattering contribution generates a very weak FT peak corresponding to a bond length of 5.40–5.44 Å. These distances between heavy scatterers agree within the error limits with the values described in the literature (U–As = 3.68 Å and U–U = 5.38 Å) [1]. A scattering contribution from copper and oxygen atoms in the interlayer [Cu(H₂O)₄]²⁺ group was not detectable because amplitude damping effects increase with increasing distances. These results demonstrate that polarization effects influence the coordination numbers but do not significantly affect the distances measured by EXAFS. Thus, within the typical error limits the preferred orientation does not limit the determination of distances.

3.1.2. As K-edge

EXAFS measurements with As as absorbing atom are shown in Fig. 2. Due to the [AsO₄] tetrahedral coordination geometry, polarization effects do not occur. The arsenate tetrahedra have an As–O_{eq} bond length of 1.68 Å. The As–U distance is 3.683–.70 Å. The As–O_{eq} distance given in [1] is 1.77 Å. It follows from the comparison of the EXAFS and XRD that only the bond distances to the surrounding oxygen atoms differ significantly whereas the

Edge, β	Shell	R _{EXAFS} [Å] ^a	N ^b	σ^2 [Å ²]	ΔE_0 [eV]	Error	R _{XRD} [Å]
U L _{III} , 0°	U–O _{ax}	1.77	1.6(1)	0.0021	0.1	0.23	1.94, 1.78
	U–O _{eq}	2.29	4.5(1)	0.0035			
	U–As	3.70	2.7(2)	0.0039			
	U–U	5.44	2.2(7)	0.0079			
U L _{III} , 45°	U–O _{ax}	1.79	2.2(1)	0.0024	1.1	0.19	
	U–O _{eq}	2.29	3.2(1)	0.0028			
	U–As	3.69	2.0(2)	0.0035			
	U–U	5.40	1.9(6)	0.008 ^c			
As K, 0°	As–O _{eq}	1.68	5.1(1)	0.0025	-7.2	1.10	1.77
	As–U	3.68	2.8(3)	0.0042			
As K, 45°	As–O _{eq}	1.68	5.0(1)	0.0022	-5.9	0.92	
	As–U	3.70	1.4(2)	0.0025			
Cu K, 45°	Cu–O	1.95	2.9(1)	0.0029	-5.2	0.58	2.14

Table 2. EXAFS structural parameters for the copper uranyl arsenate sample **1**. For comparison the atomic bond distances R_{XRD} from literature [1], at 298 K.

a: Errors in distances R are ± 0.02 Å

b: Errors in coordination numbers N are $\pm 25\%$ with standard deviations in parentheses, σ^2 – Debye-Waller factor, β – angle between polarization vector ϵ and sample surface (see Fig. 1).

c: Value fixed during the fit

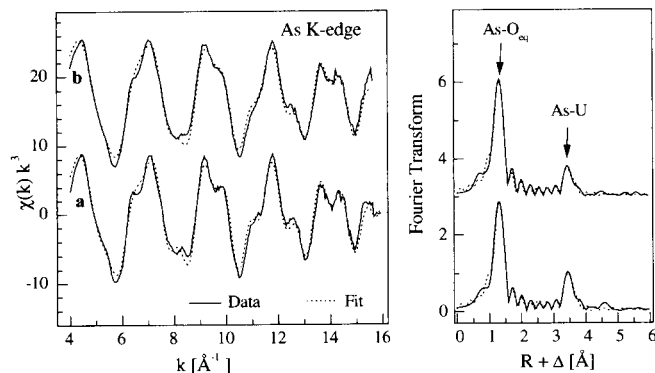


Fig. 2. As K-edge k^3 -weighted EXAFS spectra of copper uranyl arsenate sample **1** (left) and the corresponding Fourier transforms (right) with $\beta = 0^\circ$ (a) and $\beta = 45^\circ$ (b) measured at $T = 298$ K.

As–U distances are in agreement. The distances to the heavy scatterers in meta-zeunerite were the subject of a more detailed study using low-temperature EXAFS measurements at the U L_{III} - and As K-edges [19]. Both the measurements at the U L_{III} -edge and at the As K-edge detected atomic distances within the $[\text{UO}_2\text{AsO}_4]_\infty$ layer only.

3.1.3. Cu K-edge

The Cu–O distance was determined from Cu K-edge EXAFS (Fig. 3). Merely one peak is visible in the FT at room temperature. To calculate Cu–O phase and amplitude functions, a simple one-shell cluster was built using the atomic coordinates for the $[\text{Cu}(\text{H}_2\text{O})_4]^{2+}$ group in meta-torbernite [20]. For the copper uranyl arsenate sample **1**, the least square fit gave an average Cu–O bond length of 1.95 \AA . This value reflects predominantly the scattering contributions from the coordinated water in the $[\text{Cu}(\text{H}_2\text{O})_4]^{2+}$ group. No contribution of the O_{ax} atoms, which are expected at 2.5 \AA , could be observed in the FT. The Cu–O distance in the $[\text{Cu}(\text{H}_2\text{O})_4]^{2+}$ group measured by XRD is 2.14 \AA [1]. For comparison, the Cu–O bond length in the isostructural mineral meta-torbernite, $\text{Cu}[\text{UO}_2\text{PO}_4]_2 \cdot 8 \text{H}_2\text{O}$, is 1.92 \AA [20].

Low-temperature Cu K-edge EXAFS measurements at 15 K were performed to study the main structural difference between $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$ and $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 8 \text{H}_2\text{O}$. Sample **1** was representative for $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 8 \text{H}_2\text{O}$ and sample **3** for $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$, respectively. Cu

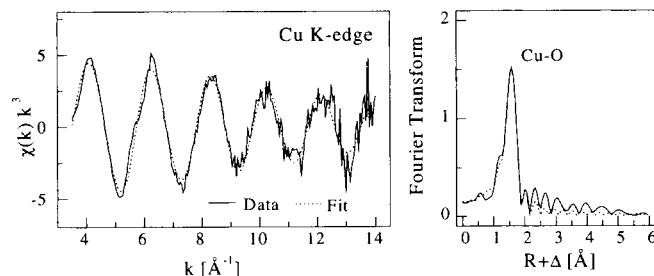


Fig. 3. Cu K-edge k^3 -weighted EXAFS spectra of copper uranyl arsenate sample **1** (left) and the corresponding Fourier transform (right) with $\beta = 45^\circ$ measured at $T = 298$ K.

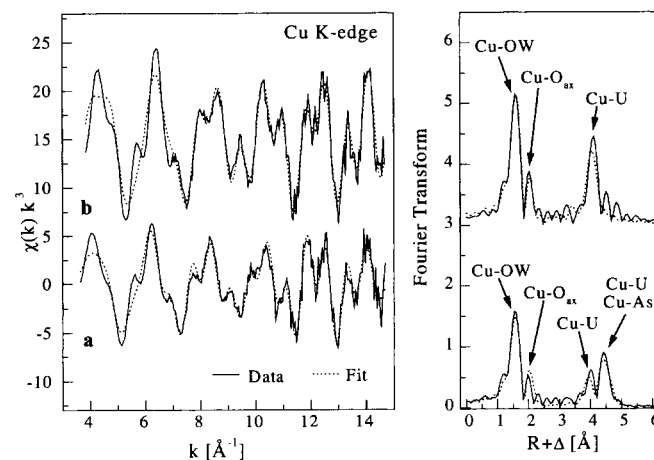


Fig. 4. Cu K-edge k^3 -weighted EXAFS spectra (left) and the corresponding Fourier transforms (right) with $\beta = 45^\circ$ of sample **1** (a) and sample **3** (b) measured at $T = 15$ K.

K-edge EXAFS spectra at low temperature are shown in Fig. 4 and the fit results are given in Table 3. Due to the damping of thermal oscillations at 15 K, additional backscattering shells occur in the FT (compare Figs. 3 and 4). To simplify the data analysis, the FT between $R + \Delta = 5.5\text{--}10 \text{\AA}$ was Fourier filtered, back transformed and subtracted from raw EXAFS data. The $[\text{Cu}(\text{H}_2\text{O})_4]^{2+}$ group causes dominant FT peaks with Cu–O distances of 1.94 \AA for both samples **1** and **3**. The scattering contribution of O_{ax} gives a weak peak at a distance of 2.46 \AA . Sample **3** shows one Cu–U peak at a distance of 4.22 \AA . A strong Cu–U– O_{ax} –Cu MS contribution appears in the FT of sample **3** due to the linear arrangement of the corresponding atoms. This observation points

Table 3. Cu K-edge EXAFS structural parameters for sample **1**, meta-zeunerite, $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 8 \text{H}_2\text{O}$ and sample **3**, zeunerite, $\text{Cu}[\text{UO}_2\text{AsO}_4]_2 \cdot 12 \text{H}_2\text{O}$, at 15 K.

Edge, β	Shell	R_{EXAFS} [\AA] ^a	N^b	σ^2 [\AA^2]	ΔE_0 [eV]
Cu K, 45° sample 1	Cu–OW _{CuI}	1.94	2.4(1)	0.0018	–13.9
	Cu–O _{ax}	2.46	1.2(1) ^d	0.0018 ^d	
	Cu–U'	4.04	0.8(1)	0.002 ^c	
	Cu–U''	4.52	0.7(1) ^c	0.002 ^c	
	Cu–As	4.84	1.6(2)	0.002 ^c	
Cu K, 45° sample 3	Cu–OW _{CuI}	1.94	3.1(2)	0.0016	–4.4
	Cu–O _{ax}	2.46	1.1(2)	0.0016 ^d	
	Cu–U	4.22	1.3(5)	0.0013	
	Cu–U _{MS}	4.22 ^d	2.7 ^d	0.0026 ^d	

a: Errors in distances R are $\pm 0.02 \text{\AA}$

b: Errors in coordination numbers N are $\pm 25\%$ with standard deviations in parentheses, σ^2 – Debye-Waller factor, β – angle between polarization vector ε and sample surface (see Fig. 1).

c: Value fixed during the fit.

d: Linked during the least-square refinement to the previous variable.

